

# ME571/Geol571 Geology and Economics of Strategic and Critical Minerals

Virginia T. McLemore

## Commodities: Introduction

# ASSIGNMENT

- **NEXT WEEK (Feb 12) IS Be, REE AFTER THAT (Feb 19, March 5, 12)**
- Barton and Young, S., 2002, Non-pegmatitic deposits of beryllium: mineralogy, geology, phase equilibria and origin: Reviews in Mineralogy and Geochemistry, v. 50, p. 591-691.  
[http://www.geo.arizona.edu/~mdbarton/MDB\\_papers\\_pdf/Barton02\\_BeRiMG050.pdf](http://www.geo.arizona.edu/~mdbarton/MDB_papers_pdf/Barton02_BeRiMG050.pdf)
- McLemore, 2010, NMBGMR OF 533,  
[http://geoinfo.nmt.edu/publications/openfile/downloads/OFR500-599/526-550/533/ofr\\_533.pdf](http://geoinfo.nmt.edu/publications/openfile/downloads/OFR500-599/526-550/533/ofr_533.pdf)

# No class Feb 26—SME

- Midterm exam will be handed out next week—due March 19 (e-mail)
- March 5—each of you will discuss SME or assigned papers
- Those who go to SME
  - Summarize 2 talks in the industrial minerals or mining & exploration sessions on REE, strategic minerals, industrial minerals, exploration (1 page summary to hand in and verbal summary)
  - My talk is on Tues at 2:00
- Those who stay home
  - Assigned articles on cathodoluminescence or REE mineralogy (1 pg summary, verbal summary)

- What products use critical and strategic minerals?
- Why are critical and strategic minerals important?
- What are some challenges in producing these minerals?

# The Changes in American Lifestyle

*1776 vs. 2005*

## *The Minerals We Use In Our Everyday Lives*

*From the Mineral Information Institute*

*An Affiliate of the Society for Mining, Metallurgy and Exploration Foundation*

*Sources: U.S. Bureau of Mines, U.S. Geological Survey, Statistical Abstracts of the USA*



[www.mii.org](http://www.mii.org)



# In 1776

Average Life span was about 33 years  
Population was 2.8 million



At the time of the Revolutionary War, about 1,200 pounds of minerals were needed each year, for every person in the United States.

— 1975, U.S. Bureau of Mines

Cement (lime)	12	Sand, gravel & stone	1,000
Clay	100	Lead	2
Coal	40	Potash	1
Copper	1	Sulfur	1
Pig Iron	20	Zinc	0.5
Salt	4		

**43,200 lbs. in a Lifetime**

## By 1850, life had improved (a little)

	<u>1776</u>	<u>1850</u>
Population (millions):	2.8	23
Lifespan (years):	33	43
Annual consumption mined material (lbs.):	1,200 (est.)	unknown
Houses (millions):	unknown	3.4



### Lifestyle in 1850:

In Charleston, 62% died by the age of 40 — 25% by the age of 5

Boston averaged 8.6 people per house

Energy consumption was 10 times that of 1776— 90% from wood, 10% from coal

In 1800: Boston to New York took 3 days by stage coach

## At the turn of the century, we were an emerging world power

	1776	1850	1900
Population (millions):	2.8	23	76
Lifespan (years):	33	43	47.3
Annual consumption mined material (lbs.):	1,200 (est.)	unknown	7,714
Houses (millions):	unknown	3.4	16
Motor Vehicles	—	—	8,000

### Lifestyle in 1900:

1.5 million telephone instruments

Avg. per capita consumption of liquors  
was 17.68 gallons

Dozen eggs cost 14¢



2.8 million miles of Public Road (1923)



## The Glorious 50s

	1776	1850	1900	1950
Population (millions):	2.8	23	76	152
Lifespan (years):	33	43	47.3	68.2
Annual consumption mined material (lbs.):	1,200 (est.)	unknown	7,714	25,938
Houses (millions):	unknown	3.4	16	42.5
Motor Vehicles	—	—	8,000	149 million



### Lifestyle in 1950:

Houses: 983 sq. ft., averaged 4.6 rooms      5% had no electricity  
71% had inside toilet — 15% had no kitchen sink  
96% had a radio — 12% had a TV  
50% had central heating — 1/2 with coal, 1/4 each fuel oil or gas  
53% no garage/carport  
Average Value= \$10,800    Avg. Monthly Rent= \$46  
Average Income was \$2,992    Bread cost 14¢ a loaf

## 1776 vs. 2005

	<u>1776</u>	<u>1850</u>	<u>1900</u>	<u>1950</u>	<u>2005</u>
Population (millions):	2.8	23	76	152	295
Lifespan (years):	33	43	47.3	68.2	77.8
Annual consumption mined material (lbs.):	1,200 (est.)	unknown	7,714	25,938	47,502
Houses (millions):		3.4	16	42.5	112
Motor Vehicles	--	--	8,000	149 million	237 million

### Lifestyle in 2005:

New Houses averaged: 2,400 sq. ft., 88% with 3 or more bedrooms

95% with more than 2 bathrooms

76% had central air conditioning

Average Value= \$258,000

2 million new units were built

Nearly 240 million motor vehicles and 4 million miles of roads (2.6 million hard-surfaced).

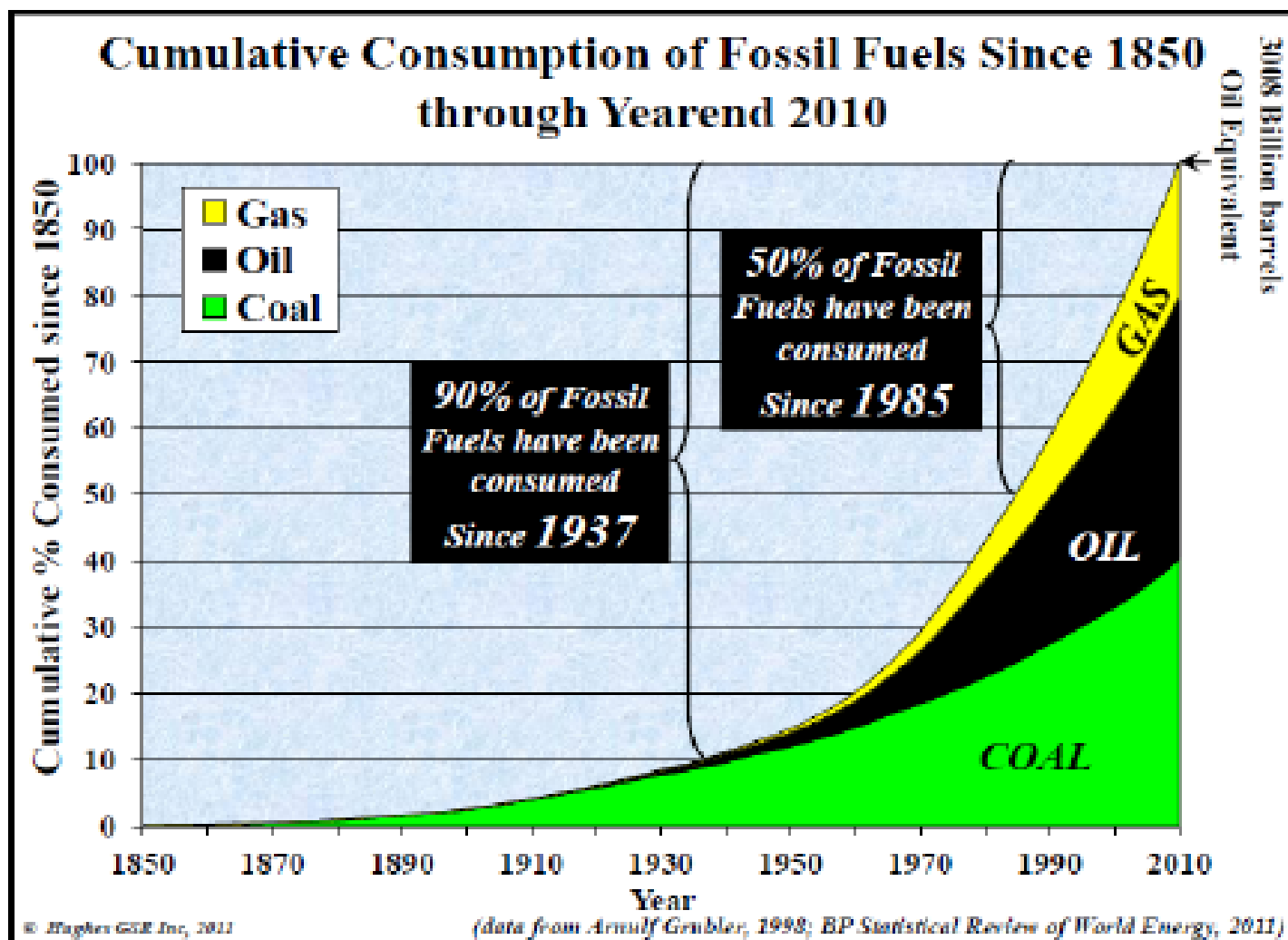
18,500 airports, serving 281,000 non-military airplanes, flying 5.5 billion miles in 8 million trips for 550 million passengers.

62% of homes owned a computer — the Internet and computers used up to 2 percent of total U.S. electricity consumption. Ordering a book on-line burned an ounce of coal.

This change in life style results in new products, which depends upon more and new minerals!

These minerals will be mined from traditional and new countries and types of deposits.

# Fossil Fuels – a similar picture to mineral resources

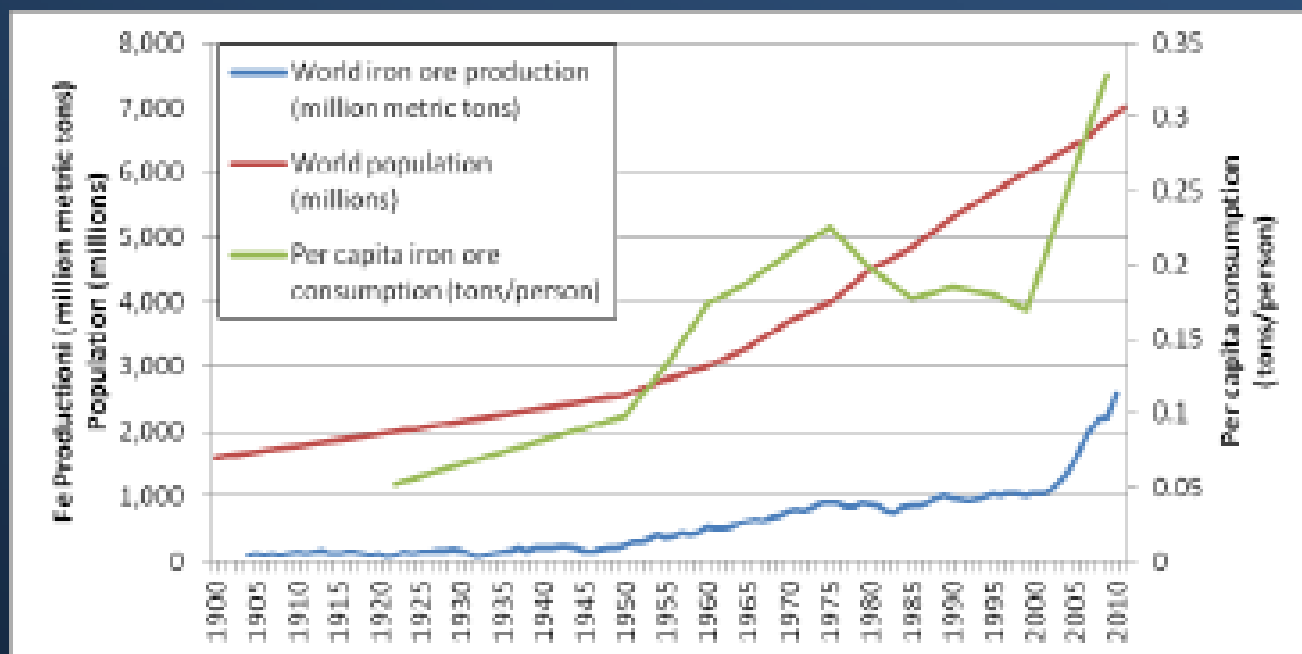


J. David Hughes, 2012

(<http://www.eeb.cornell.edu/howarth/HUGHES%20Cornell%20Ithaca%20May%202%202012.pdf>)

# Mineral Resources – The Big Picture

## Global Trends in Population, Iron Ore Production, & Consumption, 1990-2011



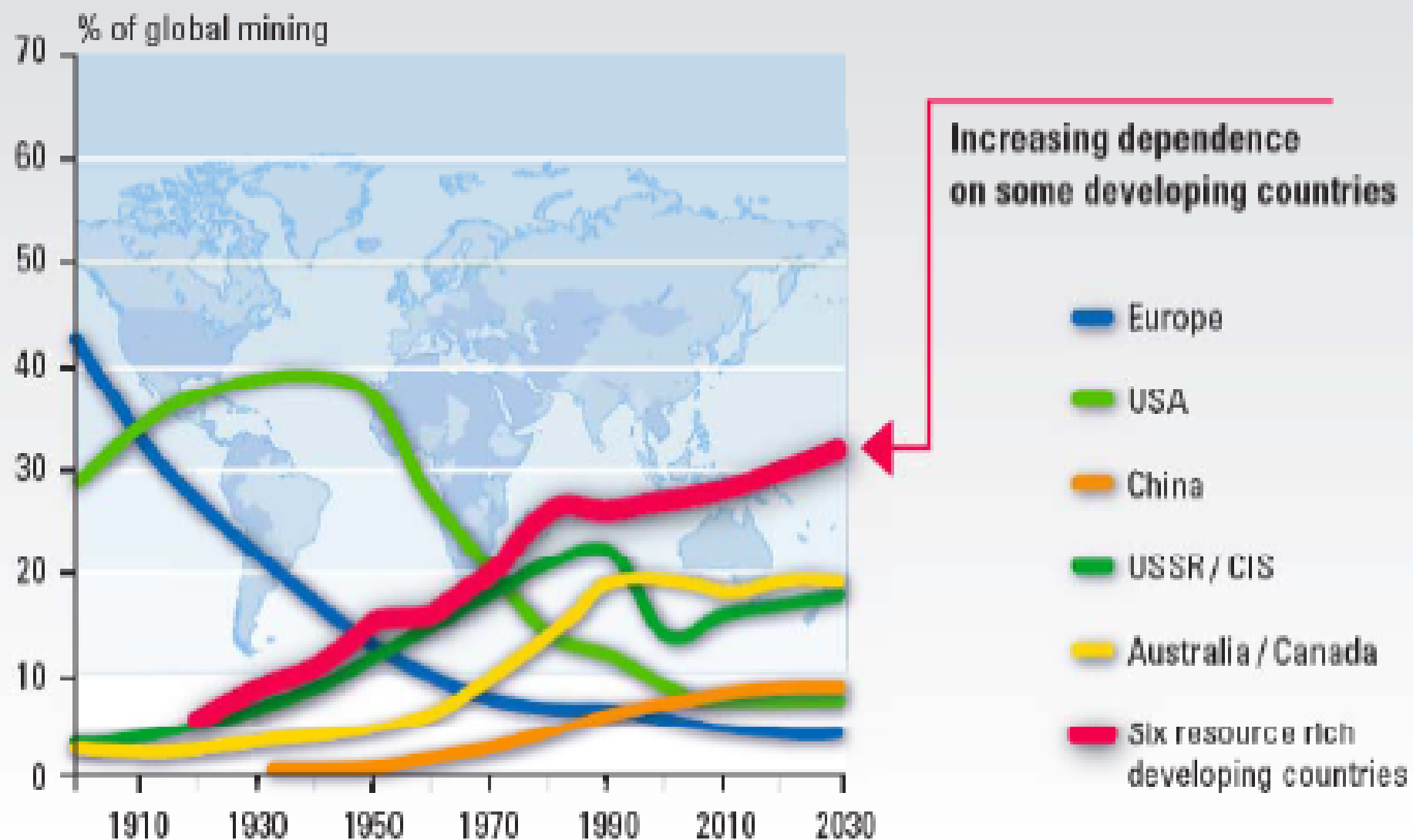
Production data from USGS

~4X more population than 100 years ago

~6X more per capita iron consumption than 100 years ago

~26X more iron ore production than 100 years ago

## World Mining 1900 - 2030



Sources: Ericsson, Raw Materials Data 2005; Sames, 1975

# Various applications of Critical and strategic minerals



## Usage of significant metals in 2030 by SET-Plan technology (% of 2010 world supply)

Metal	Solar PV	Wind	Nuclear	CCS	Total
Tellurium	50.4%	✕	✕	✕	50.4%
Indium	18.0%	✕	1.4%	✕	19.4%
Tin	9.6%	✕	0.02%	✕	9.6%
Hafnium	✕	✕	7.0%	✕	7.0%
Silver	4.8%	✕	0.4%	✕	5.2%
Dysprosium	✕	4.0%	✕	✕	4.0%
Gallium	3.9%	✕	✕	✕	3.9%
Neodymium	✕	3.8%	✕	✕	3.8%
Cadmium	1.5%	✕	0.03%	✕	1.5%
Nickel	✕	0.7%	0.2%	0.5%	1.5%
Molybdenum	✕	1.0%	0.4%	0.02%	1.4%
Vanadium	✕	✕	0.01%	1.3%	1.3%
Niobium	✕	✕	0.04%	1.2%	1.2%
Selenium	0.8%	✕	✕	✕	0.8%

None for bioenergy and electricity grid

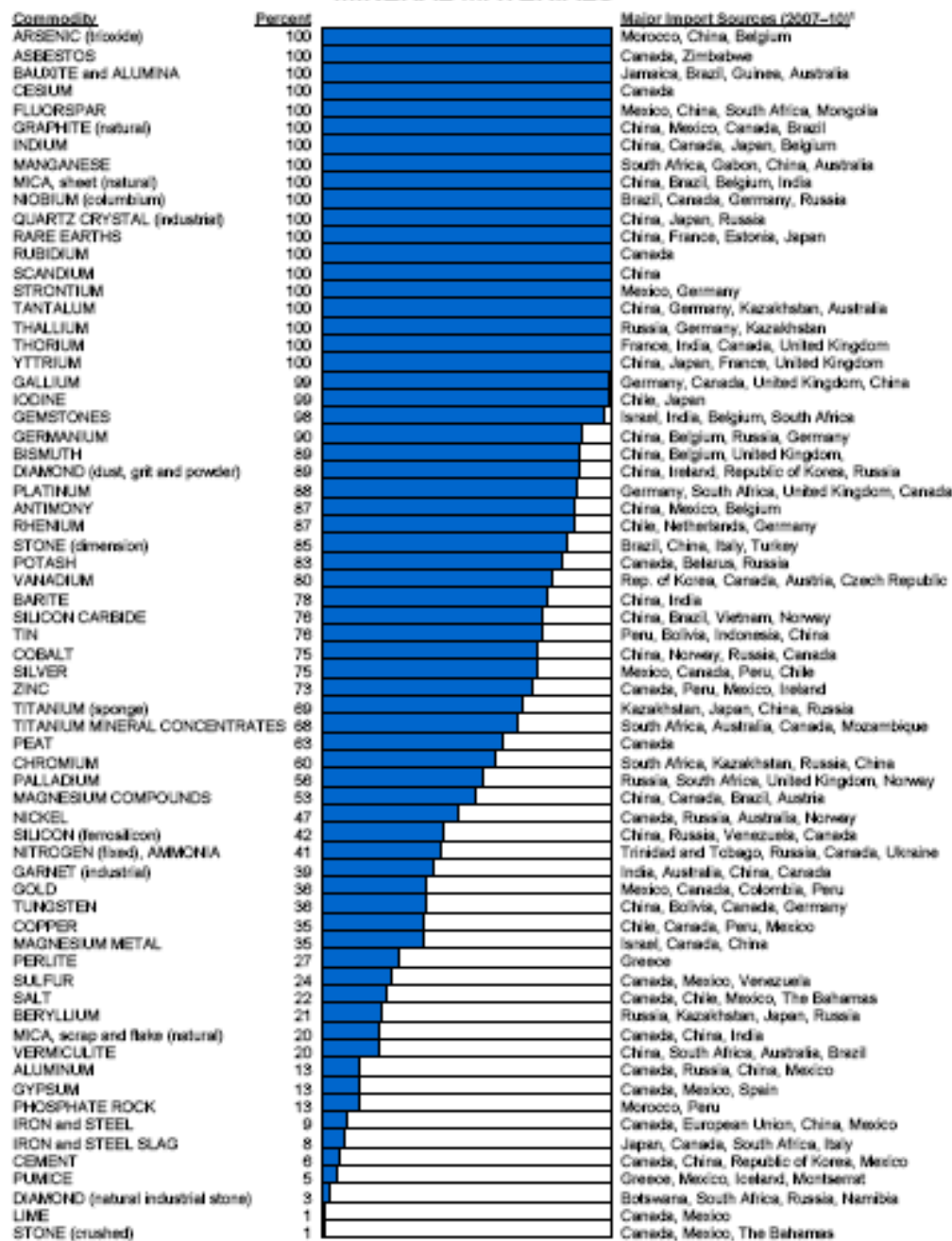


The US is 80% dependent for supply of 31 minerals.

This dependency implies criticality.

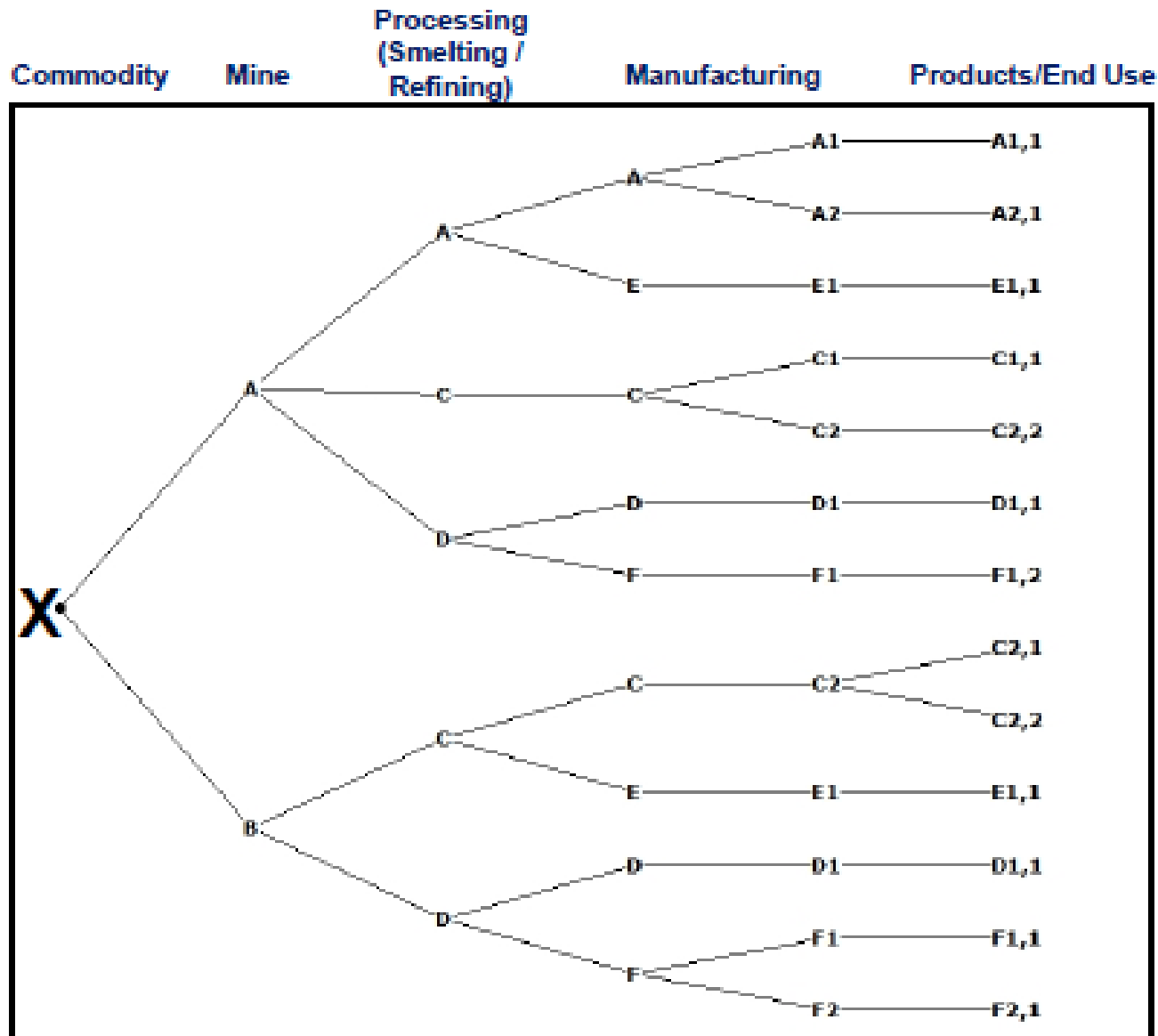
6

## 2011 U.S. NET IMPORT RELIANCE FOR SELECTED NONFUEL MINERAL MATERIALS



<sup>a</sup>In descending order of import share.

# Supply Chain Analysis



## Example

- Two sources of the commodity X (A,B)
- Three facilities that can process the raw material (A,C,D)
- 4 manufactured goods: (1,1), (1,2), (2,1), and (2,2)
- Good (1,1) is produced by 7 supply chains
- Good (1,2) is produced by only one supply chain
- Each supply chain can be disrupted in multiple places
- There are multiple probabilities of disruption that must be considered, based upon both natural and man-made events

*What defines a supply disruption?*

# EMERGING TECHNOLOGIES

# Cell phones

## FACT SHEET

## CELL PHONE

### What's in my Cell Phone?

- **Arsenic (gallium arsenide in the amplifier and receiver).** Mined in China, Chile, Morocco, Peru, Kazakhstan, Russia, Belgium and Mexico.
- **Copper (circuitry).** Mined in Chile, United States, Peru, China, Australia, Russia, Indonesia, Canada, Zambia, Poland, Kazakhstan and Mexico.
- **Gallium (gallium arsenide).** Mined in China, Germany, Kazakhstan and Ukraine.
- **Gold (circuitry).** Mined in China, United States, Australia, South Africa, Peru, Russia, Canada, Uzbekistan, Ghana, Papua New Guinea, Indonesia, Brazil, Mexico and Chile.
- **Magnesium compounds (phone case).** Mined in China, Turkey, North Korea, Russia, Slovakia, Austria, Spain, Australia, Brazil, Greece, India and the United States.
- **Palladium (circuitry).** Mined in Russia, South Africa, Canada, United States and Zimbabwe.
- **Platinum (circuitry).** Mined in South Africa, Russia, Canada, Zimbabwe, United States and Colombia.
- **Silver (circuitry).** Mined in Peru, Mexico, China, Australia, Chile, Russia, United States, Poland, Bolivia and Canada.
- **Tungsten (circuitry).** Mined in China, Russia, Canada, Austria, Bolivia and Portugal.
- A multitude of petroleum products are used in cellular phones.



### INTERESTING FACTS

- About 130 million cell phones are retired annually in the United States. Collectively, these cell phones weigh about 14,000 metric tons. Annually retired cell phones contain almost 2,100 metric tons of copper, 46 metric tons of silver, 3.9 metric tons of gold, 2 metric tons of palladium, and 0.04 metric tons of platinum.
- Recovery and recycling of cell phones are in the early stages of development, as is the case for recycling of electronics in general. For cell phone recycling to grow, recycling must become economically viable. Efficient recovery infrastructure, product designs that simplify dismantling, and other changes are needed to facilitate the growth of cell phone recycling.
- Gallium arsenide is used in the amplifier and receiver.
- Magnesium compounds are alloyed to make the cell phone cases.

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www.msi.org

  
Society for  
Mining, Metallurgy  
& Exploration  
www.smenet.org

- **Cell phone performance dependence on a wide variety of minerals**
  - Scarce or expensive to process
  - **Barium titanate ceramics**: dielectric properties to avoid signal broadening and heat buildup
  - **Rare earth elements (REs)** and **indium**: ceramic magnetic switches
  - **Titanium dioxide**: dielectric heart of the phone
  - **Indium tin oxide**: liquid crystal display
  - **Tantalum**: dielectric resonators in 2.2 gigahertz cellular base stations
- **Markets for specialized minerals or mineral products in cell phone are small**
  - Volume of material needed for cell phones is small
  - Increased demand means sharp prices rises
    - **Indium**: 300\$/kg late 90`s to 1000\$/kg today



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## The Life Cycle of a Cell Phone

**Y**ou have one, your parents have one, your friends each have one—owning a cell phone has become as common as having a toothbrush in your home. More than 156 million Americans now use cell phones—including about 20 percent of American teens! In a way, cell phones have become a necessity of everyday life—we use them to call home when we're late, make plans with friends, or get directions when we're lost. But have you ever thought about how cell phones are made and what happens to them when you don't need them anymore?

Like any product, making a cell phone and its parts uses natural resources and energy, which can potentially impact the air, land, and water. Understanding the life cycle of a product can help you make environmental choices about the products you use, and how you dispose of them. You can help minimize your environmental impact of using a cell phone by:

**• Keeping your phone longer.** Choose your cell phone service provider carefully. Pick a phone with features you need and a style you like so you will keep it longer.

**• Charging your battery correctly.** Increase the life span of your phone and battery by following the manufacturer's directions for charging the battery.

**• Reusing or recycling your phone.** Find ways to reuse or recycle your phone and accessories when you're finished with them. Many companies recycle or reuse cell phones—visit the "Resources" section of this poster for a list of suggestions.

Follow the life-cycle diagram to learn more about cell phones, their parts, and their potential impact on the environment.

### Materials Extraction

A cell phone is made up of many materials. In general, the handset consists of 40 percent metals, 40 percent plastics, and 20 percent ceramics and trace materials.

• The **circuit board** (also called a printed wiring board, located in the handset) is the "brain" of the cell phone because it controls all of its functions.

• Circuit boards are made from mined, raw materials including copper, gold, lead, nickel, zinc, beryllium, barium, cobalt, and other metals. The manufacturing of these boards requires crude oil for plastic, and sand and limestone for fiberglass. Many of these materials are known as "persistent toxins" and can stay in the environment for long periods of time, even after disposal.

• The **liquid crystal display (LCD)** is a low-power, flat-panel display on the front of your phone that shows information and images. It becomes opaque (hard to see through) when an electric current passes through it. The contrast between the opaque and transparent (see-through) areas forms visible characters.

*Approximately 10 percent of America's total power (for the last 10 years) is used to produce the LCDs in cell phones. By 2010, the use of LCDs in cell phones will be 10 percent of total power.*

• Various liquid crystalline substances, either naturally occurring (such as mercury, a potentially dangerous substance) or human-made, are used to make LCDs. LCDs also require the use of glass or plastic.

• The **rechargeable battery** is used to power the phone.

• Cell phones can use several types of batteries: nickel metal hydride (NiMH), lithium-ion (Li-ion), nickel cadmium (NiCd), or lead acid. NiMH and NiCd batteries contain nickel, cobalt, zinc, cadmium, and copper. Li-ion batteries use lithium-metal oxide and carbon-based materials, all mined from the earth.

### Materials Processing

Most raw materials must be processed before manufacturers can use them to make products. For example, in cell phones:

• Crude oil is combined with natural gas and chemicals in a processing plant to make plastic.

• Copper is mined, ground, balled, and treated with chemicals and electricity to make the pure metal used to make circuit boards and batteries. The resulting copper pieces are shipped to a manufacturer where they are formed into wires and sheets.

### Manufacturing

Plastics and fiberglass are used to make the basic shape of the **circuit board**, which is then coated with gold plating. The board is also composed of several electronic components, connected with circuits and wires (primarily made of copper) that are soldered to the board and secured with protective glass and coatings.

LCDs are manufactured by sandwiching liquid crystal between layers of glass or plastic.

Batteries consist of two separate parts, called electrodes, made from two different metals. A liquid substance, called electrolyte, touches each electrode. When an outside source of electricity such as an outlet is applied, chemical reactions between the electrodes and the electrolyte cause an electric current to flow, giving batteries their "juice" or power.

*By 2010, cell phones will be responsible for a rise of more than 100 million pounds of plastic, resulting in more than 100 million pounds of waste.*

### Using Less Stuff

Cell phone companies have made great strides in "dematerialization" using less material as shown by the decreasing size of today's cell phones. Years ago, the technology needed for a cell phone would have filled the entire floor of an office building; now everything needed for a cell phone weighs only 7.7 ounces!

## The Life Cycle of a Cell Phone



Disposal

Materials Extraction

Materials Processing

Recycling

Manufacturing

End-of-Life

Reuse

Useful Life

Packaging & Transportation



### The Nine Lives of a Cell Phone

Cell phones consist of nine basic parts, each of which has its own life cycle:

- Keypad
- Microphone
- Speaker
- Plastic casing
- Accessories (such as adapters, headsets, carrying cases, and decorative face plates)
- Battery
- Antenna
- Circuit board/printed wiring board
- Liquid crystal display (LCD)

### Packaging & Transportation

Cell phone parts and the finished products need packaging and transportation to get from one place to another. Transportation by plane, truck, or rail all require the use of fossil fuels for energy, which can contribute to global climate change.

While packaging protects products from damage, excessive or decorative packaging can be wasteful. Packaging consumes valuable natural resources, such as paper (from trees), plastic (from crude oil in the earth), and metals (from ore), or other materials, all of which use energy to produce and can result in waste. Some packaging, however, can be made from recycled materials.

### Useful Life

Unlike other countries, cell phone companies in the United States sell their own phones, which are usually not interchangeable from company to company. Even though regulations now allow consumers to transfer their phone number to a new phone company, most companies have unique technologies in their phones that only work in their own networks. This means that switching cell phone companies can mean having to purchase a new phone. One way to extend the useful life of your phone and prevent waste is to use the same company for continuing phone service. Always comparison shop to be sure you get the service and phone that's right for you.

You can also extend the life of your phone by taking care of it—protecting it from damage by storing it in a case, avoiding overheating it, and keeping it out of extreme heat and cold and away from water and other liquids.

The use of **rechargeable batteries** in cell phones reduces the amount of waste and toxicity that disposable batteries create. Be sure to follow the manufacturer's instructions for charging your batteries so you can extend their lives as long as possible.

### Reuse

Many organizations—including recyclers, charities, and electronics manufacturers—accept working cell phones and offer them to schools, community organizations, and individuals in need. Reuse gives people, who could not otherwise afford them, free or reduced cost access to new phones and their accessories. Plus, it extends the useful lifetime of a phone.

### End-of-Life

Donating or recycling cell phones when you no longer need or want them extends their useful lives, and prevents them from ending up in the trash where they can potentially cause environmental problems.

**Reuse**  
Many organizations—including recyclers, charities, and electronics manufacturers—accept working cell phones and offer them to schools, community organizations, and individuals in need. Reuse gives people, who could not otherwise afford them, free or reduced cost access to new phones and their accessories. Plus, it extends the useful lifetime of a phone.

### Take-Back Programs

Many cell phone manufacturers and service providers offer a "take-back" program. Under this system, manufacturers accept used cell phones and accessories and either recycle, re-manufacture, or dispose of them using systems designed to handle the specific types of waste cell phones produce. Contact your manufacturer by using the information that came with your phone or via the Internet.



### Recycle

Electronics recyclers are springing up everywhere! Today, many stores, manufacturers, and recycling centers accept cell phones for recycling. While some electronics recyclers only accept large shipments, communities, schools, or groups can work together to collect used cell phones for shipment to electronics recyclers.

Some rechargeable batteries can also be recycled, as several retail stores and some communities have started collecting them. When rechargeable batteries are recycled, the necessary materials can be used to make new batteries and stainless steel products.

Check the "Resources" section of this poster for a list of organizations that will accept your phone and accessories for reuse or recycling. You can also use the Internet or phone book to search for local contacts that recycle and refurbish cell phones.

### Dispose

By 2005, the rate at which cell phones are discarded is predicted to exceed 125 million phones each year, resulting in more than 65,000 tons of waste! Cell phones that are thrown in the trash end up in landfills (buried in the ground) or incineration (burned). Because cell phones contain metals, plastics, chemicals, and other potentially hazardous substances, you should always recycle, donate, or trade in your old cell phone. It's free and easy. Don't throw it away! Phones that are thrown away waste energy and result in the loss of valuable resources.

### Crank up the Volume

A major cell phone manufacturer recently developed a way to recharge cell phone batteries using "muscle power." This hand-powered device provides 20 minutes of talk time after just three minutes of squeezing a hand-held generator! Other new technologies, such as hydrogen fuel cells and solar-charged batteries, are under development and might ultimately replace current battery technology. These new alternatives will conserve natural resources and reduce waste.

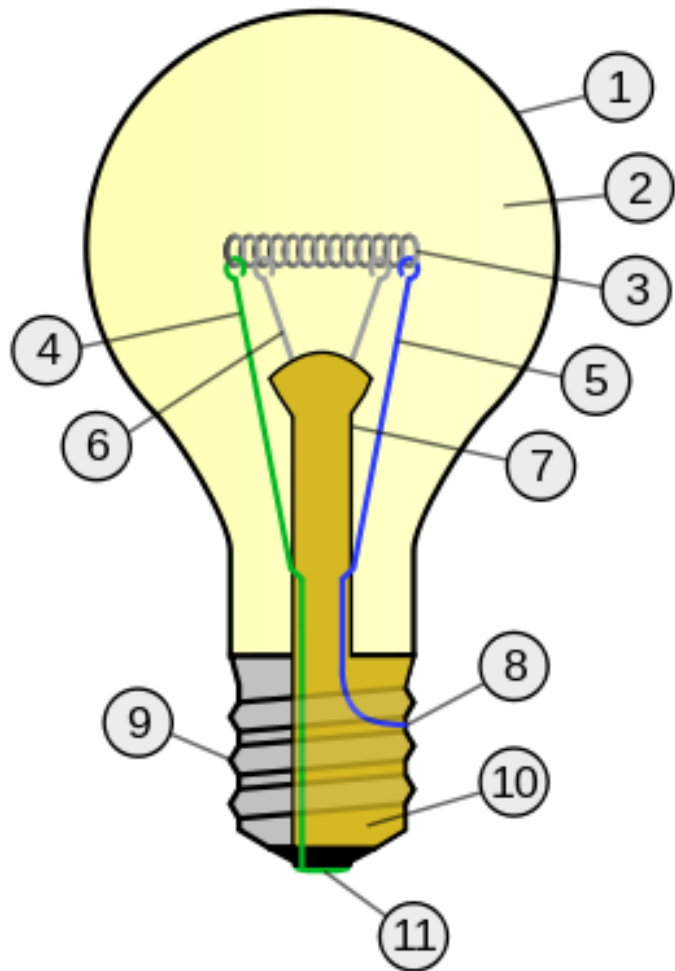


*Between 1999 and 2003, 125 million pounds of plastic were used to make cell phones. By 2010, the use of plastic in cell phones will be 100 million pounds of plastic.*

<http://www.epa.gov/osw/education/pdfs/life-cell.pdf>

# LIGHT BULBS





1. Outline of Glass bulb
2. Low pressure inert gas (argon, nitrogen, krypton, xenon)
3. Tungsten filament
4. Contact wire (goes out of stem)
5. Contact wire (goes into stem)
6. Support wires (one end embedded in stem; conduct no current)
7. Stem (glass mount)
8. Contact wire (goes out of stem)
9. Cap (sleeve)
10. Insulation ([vitrite](#))
11. Electrical contact

## What's in a Compact Fluorescent Light Bulb?



- **Barite (for phosphor).** Mined in China, India, United States, Morocco, Iran, Turkey, Mexico, Kazakhstan, Vietnam, Germany, Russia, Algeria, United Kingdom and Pakistan.
- **Bauxite (alumina for phosphor; aluminum for end caps & filaments).** Mined in Australia, China, Brazil, India, Guinea, Jamaica, Russia, Venezuela, Suriname, Kazakhstan, Guyana and Greece.
- **Copper (end caps; filaments).** Mined in Chile, United States, Peru, China, Australia, Russia, Indonesia, Canada, Zambia, Poland and Mexico.
- **Lead (soda-lime glass; ballast; adapter unit).** Mined in China, Australia, United States, Peru, Mexico, Canada, India, Bolivia, Poland, Russia, Sweden, Ireland and South Africa.
- **Limestone or Dolomite (finely-crushed stone to make soda-lime glass).** Mined in United States.
- **Mercury (vapor in glass tubing).** Mined in China, Kyrgyzstan and Peru.
- **Nickel (end caps; filaments).** Mined in Russia, Canada, Australia, Indonesia, New Caledonia, Philippines, Columbia, China, Cuba, Brazil, Botswana, South Africa, Dominican Republic, Greece, Venezuela and Spain.
- **Phosphate Rock (phosphor).** Mined in China, United States, Morocco & Western Sahara, Russia, Tunisia, Jordan, Brazil, Syria, Israel, Egypt, Australia, South Africa and Canada.
- **Rare Earth Oxides (Lanthanum or Yttrium for phosphor).** Mined in China, India and Brazil.
- **Silica (glass).** Mined in United States, Italy, Germany, United Kingdom, Australia, France, Spain, Japan, Poland, Hungary, South Africa, Mexico, Austria, Iran, Republic of Korea, Slovakia, Canada, Belgium, India, Bulgaria, Norway, Chile, Gambia, Turkey and Czech Republic.
- **Soda Ash (soda-lime glass).** Mined in United States, Kenya and Botswana.
- **Manganese (phosphor).** Mined in South Africa, Australia, China, Gabon, Brazil, India, Ukraine and Mexico.
- **Tin (end caps; filaments; glass coatings).** Mined in China, Indonesia, Peru, Bolivia, Brazil, Congo-Kinshasa, Vietnam, Malaysia, Australia and Russia.
- **Tungsten (electrodes; filaments).** Mined in China, Russia, Canada, Austria, Bolivia and Portugal.
- **Zinc (end caps; filaments).** Mined in China, Peru, Australia, United States, Canada, India, Kazakhstan, Ireland and Mexico.



To learn more about minerals and mining visit  
[www.MineralsEducationCoalition.org](http://www.MineralsEducationCoalition.org)

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## INTERESTING FACTS

- CFLs are known as *compact fluorescent lights* or *compact fluorescent light bulbs*. In a CFL, an electric current is driven through a glass tube containing argon and a small amount of mercury vapor. This generates invisible ultraviolet light that excites a fluorescent coating (called phosphor) on the inside of the tube, which then emits visible light.
- CFLs are made of soda-lime glass, similar to that used throughout the glass industry for bottles and other common products.
- Phosphor in a CFL is a phosphate mix that may contain manganese, rare elements such as lanthanum, and yttrium as either an oxide or a phosphate, along with a barium/aluminum oxide. Phosphor components may vary slightly depending on the color of the lamp.
- While a regular (incandescent) light bulb uses heat to produce light, a fluorescent bulb creates light using an entirely different method that is 4 to 6 times more energy-efficient. This means that a 15-watt CFL produces the same amount of light as a 60-watt regular incandescent bulb. CFLs last up to 13 times longer and use 2/3 to 3/4 less electricity than incandescent bulbs with similar lumen ratings.
- CFLs contain a very small amount of mercury sealed within the glass tubing – an average of 4 milligrams. By comparison, older thermometers contain about 300 milligrams of mercury – an amount equal to the mercury in 125 CFLs. Mercury is an essential part of CFLs; it allows the bulb to be an efficient light source. No mercury is released when the bulbs are intact (not broken) or in use. Because the CFLs contain mercury, the U.S. Environmental Protection Agency encourages their recycling after they burn out. In some states, CFL recycling may be mandatory.
- China supplies 97% of the world's supply of rare earths, which are used in a variety of products.
- The U.S. possesses the largest non-China rare earth resource in the world at the Mountain Pass Mine in California.

CD/DVD

## What's That Disc Made Of?



- **Bauxite (aluminum):** Mined in Australia, China, Brazil, India, Guinea, Jamaica, Russia, Venezuela, Suriname, Kazakhstan, Guyana and Greece.
- **Gold (reflects laser beam):** Mined in China, United States, Australia, South Africa, Peru, Russia, Canada, Uzbekistan, Ghana, Papua New Guinea, Indonesia, Brazil, Mexico and Chile.
- **Oil, Oil Sands or Oil Shale (polycarbonate plastic):** From oil drilling, mined oil sands, or oil shale.

### SOURCES

<http://minerals.usgs.gov/minerals/pubs/mcs/>  
[www.mediatechnics.com/dvdcopiers.htm](http://www.mediatechnics.com/dvdcopiers.htm)  
[www.all4dvd.com/images/dvd\\_discs.png](http://www.all4dvd.com/images/dvd_discs.png)  
[www.mediatechnics.com/dvdcopiers.htm](http://www.mediatechnics.com/dvdcopiers.htm)  
[www.earth911.com](http://www.earth911.com)  
[www.cdrecyclingcenterofamerica.com](http://www.cdrecyclingcenterofamerica.com)  
[www.kodak.com](http://www.kodak.com)

### INTERESTING FACTS

- The discs are made of plastics and metals. The largest ingredient is polycarbonate plastic derived from oil. Can be processed from oil drilling or mined from oil sand or oil shale.
- The most important part of the disc is the thin metal layer that reflects laser beams used to read the information on the disc.
- Most discs are thrown into landfills because people are not aware that they can be recycled. *Goodwill*, *Salvation Army* and *Best Buy* recycle CDs and DVDs. *Back Thru the Future*, *CD Recycling Center of America* and *Green Disk* are located online and offer many tips on how to recycle CDs and DVDs.

Researcher: Ben Larson,  
Colorado School of Mines

Edited by the U.S. Geological Survey 2010



To learn more about minerals and mining visit  
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## INTERESTING FACTS

- The discs are made of plastics and metals. The largest ingredient is polycarbonate plastic derived from oil. Can be processed from oil drilling or mined from oil sand or oil shale.
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# Elements in Computer Chips (National Research Council, 2007)

elements needed in 1980s  
 additional elements needed today

H	additional elements needed today																He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac															
			Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	



# What are green technologies?



- Environmental technologies or clean technologies
- Future and existing technologies that conserve energy and natural resources and curb the negative impacts of human involvement, i.e. environmental friendly (modified from Wikipedia)
  - Alternative power (wind turbines, solar energy)
  - Hybrid and electric cars
  - Batteries
  - Magnets
- Other technologies
  - Water purification
  - Desalination
  - Carbon capture and storage





- solar panels/photovoltaics
- wind turbines
- batteries
- magnets
- other

H	other																He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac															
			Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

# Common minerals

- Cement/concrete
- Copper
- Steel
- Aluminum
- Titanium

# AUTOMOBILES

**TABLE 2.2** Some Minerals and Their Weights and Properties in Today's Automobile

Mineral	2006 Weight (pounds/kilograms)	Property
Iron & Steel	2124/ 963	High strength, durability (frame, motor)
Aluminum	240/109	Light weight (frame, motor)
Carbon	50/23	Bond strengthener (tires and other rubber parts)
Copper	42/19	Electrical conductivity
Silicon	41/19	Bonding properties (windshields and windows)
Lead	24/11	Conductor (storage batteries)
Zinc	22/10	Galvanizer, strengthens in metal alloys (die cast parts and galvanized metal)
Manganese	17/8	Hardens as metal alloy
Chromium	15/7	Corrosion resistance and hardness as metal alloy
Nickel	9/4	Strength at elevated temperature and corrosion resistance as metal alloy
Magnesium	4.5/2	Alloying element with other metals like aluminum
Sulfur	2/0.9	Strengthens rubber tires
Molybdenum	1/4.5	Strength and toughness as metal alloy
Vanadium	<1/<0.45	Strengthens, hardens, lighter weight as metal alloy
Platinum	0.05-0.10 troy ounce/ 1.5 – 3.0 grams	Catalytic properties (catalytic converters)

[http://www.nma.org/pdf/101606\\_nrc\\_study.pdf](http://www.nma.org/pdf/101606_nrc_study.pdf)

*Table 2-1. Materials in Clean Energy Technologies and Components*

CLEAN ENERGY TECHNOLOGIES AND <i>COMPONENTS</i>						
		Solar Cells	Wind Turbines	Vehicles		Lighting
MATERIAL		<i>PV films</i>	<i>Magnets</i>	<i>Magnets</i>	<i>Batteries</i>	<i>Phosphors</i>
Rare Earth Elements	Lanthanum				●	●
	Cerium				●	●
	Praseodymium		●	●	●	
	Neodymium		●	●	●	
	Samarium		●	●		
	Europium					●
	Terbium					●
	Dysprosium		●	●		
	Yttrium					●
Indium	●					
Gallium	●					
Tellurium	●					
Cobalt				●		
Lithium				●		



Toyota Prius

2.2 lbs Nd in magnets

22-33 lbs La in batteries

[http://www.molycorp.com/hybrid\\_ev.asp](http://www.molycorp.com/hybrid_ev.asp)

# BATTERIES

# Batteries

- Primary (immediate current, disposable) vs secondary (must be charged)
- Types
  - Wet cell
  - Dry cell
  - Molten salt
  - reserve
- Dependent upon use
  - Computers, cell phones
  - Solar power system



# Significant Growth in Portable Lithium-Ion Batteries

→ Mobile Electronics



*Table 2-1. Materials in Clean Energy Technologies and Components*

CLEAN ENERGY TECHNOLOGIES AND <i>COMPONENTS</i>						
		Solar Cells	Wind Turbines	Vehicles		Lighting
MATERIAL		<i>PV films</i>	<i>Magnets</i>	<i>Magnets</i>	<i>Batteries</i>	<i>Phosphors</i>
Rare Earth Elements	Lanthanum				●	●
	Cerium				●	●
	Praseodymium		●	●	●	
	Neodymium		●	●	●	
	Samarium		●	●		
	Europium					●
	Terbium					●
	Dysprosium		●	●		
	Yttrium					●
Indium	●					
Gallium	●					
Tellurium	●					
Cobalt				●		
Lithium				●		

# Other materials

- Zinc
- Carbon
- Chloride
- Manganese
- Nickel
- Lithium
- Copper
- Silver
- Cadmium
- Lead
- REE
- Iron
- mercury

# PHOTOVOLTAICS

## Metals & Mineral Products used to make a Solar Panel



- **Arsenic (gallium-arsenide semiconductor chips).** Mined in China, Chile, Morocco, Peru, Kazakhstan, Russia, Belgium and Mexico.
- **Bauxite (aluminum).** Mined in Australia, China, Brazil, India, Guinea, Jamaica, Russia, Venezuela, Suriname, Kazakhstan, Guyana and Greece.
- **Boron Minerals (semiconductor chips):** Mined in United States, Turkey, Argentina, Chile, Russia, Peru, China, Bolivia and Kazakhstan.
- **Cadmium (thin film solar cells).** Mined in China, Republic of Korea, Japan, Kazakhstan, Mexico, Canada, Russia, United States, India, Netherlands, Poland, Germany and Australia.
- **Coal (by-product coke is used to make steel).** Coal is mined world-wide, and constitutes 45% of the generation of U.S. electricity.
- **Copper (wiring; thin film solar cells):** Mined in Chile, United States, Peru, China, Australia, Russia, Indonesia, Canada, Zambia, Poland and Mexico.
- **Gallium (solar cells).** Mined in China, Germany, Kazakhstan and Ukraine.
- **Indium (solar cells).** Mined in China, Republic of Korea, Japan, Canada, Belgium, Russia, Peru and Brazil.
- **Iron ore (steel).** Mined in China, Brazil, Australia, India, Russia, Ukraine, United States, South Africa, Iran, Canada, Sweden, Kazakhstan, Venezuela and Mexico.
- **Molybdenum (photovoltaic cells).** Mined in China, United States, Chile, Peru, Mexico, Canada, Armenia, Iran, Russia and Mongolia.
- **Lead (batteries).** Mined in China, Australia, United States, Peru, Mexico, Canada, India, Bolivia, Poland, Russia, Sweden, Ireland and South Africa.
- **Phosphate rock (phosphorous).** Mined in China, United States, Morocco, Western Sahara, Russia, Tunisia, Jordan, Brazil, Syria, Israel, Egypt, South Africa and Canada.
- **Selenium (solar cells).** Mined in Japan, Belgium, Canada, Russia, Chile, the Philippines, Finland, Peru, Sweden and India.
- **Silica (solar cells).** Mined in United States, Italy, Germany, United Kingdom, Australia, France, Spain, Japan, Poland, Hungary, South Africa, Mexico, Austria, Iran, Republic of Korea, Slovakia, Canada, Belgium, India, Bulgaria, Norway, Chile, Gambia, Turkey and Czech Republic.
- **Tellurium (solar cells).** Mined in Australia, Belgium, Canada, China, Germany, Japan, Kazakhstan, Peru, Philippines, Russia and United States.
- **Titanium dioxide (solar panels).** Mined in Australia, South Africa, Canada, China, India, Norway, Ukraine, Vietnam, Mozambique, United States, Sierra Leone and Brazil.



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## INTERESTING FACTS

In 1954 Photovoltaic technology was born in the United States when Daryl Chapin, Calvin Fuller, and Gerald Pearson develop the silicon photovoltaic (PV) cell at Bell Labs—the first solar cell capable of converting enough of the sun's energy into power to run everyday electrical equipment.

Molybdenum is "sputtered" onto the photovoltaic cells as a base conductive layer for all the other layers. Sputtering is a process that uses ions of an inert gas to dislodge atoms from the surface of a crystalline material, the atoms then being electrically deposited to form an extremely thin coating on a glass, metal, plastic, or other surface.

# Photovoltaics

- Specialty
  - Commercial
    - Silicon
    - Silver
    - **Tellurium**
    - Cadmium
  - Emerging
    - **Indium**
    - **Selenium**
    - Molybdenum
    - **Gallium**
    - Germanium
    - Arsenic
    - Ruthenium
  - Many minor materials (Ni, P, Zn, Sn, S, N, H, more)
- Bulk
  - Steel
  - Aluminum
  - **Copper**
  - Adhesive
  - Insulating
  - Concrete
  - Plastic

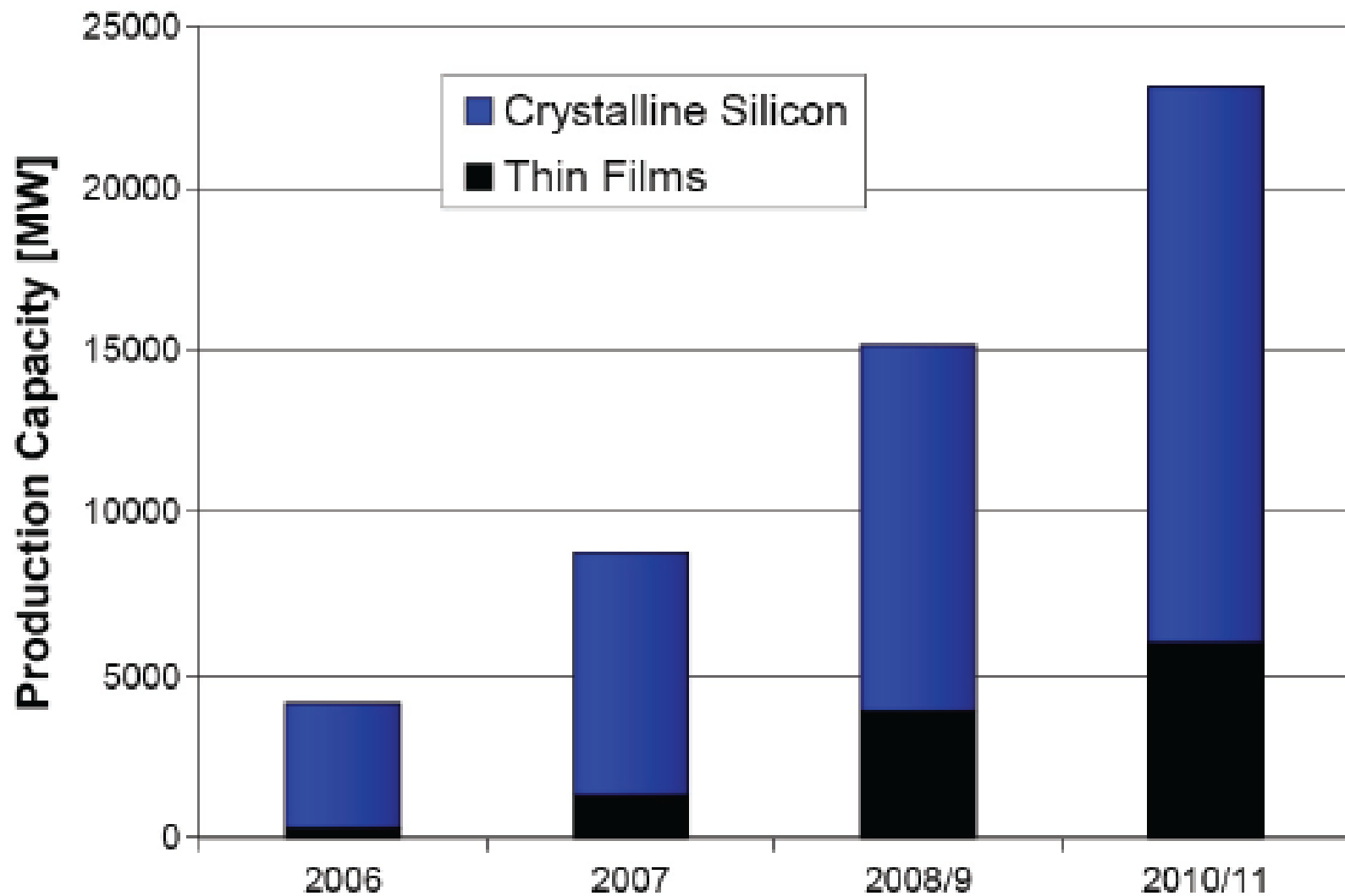


Figure 6.3: PV Production Capacities 2006 and forecast (Source: Jäger-Waldau 2007)

Figure 6.3 shows the actual and planned production capacities for the worldwide photovoltaic industry. Clearly visible is the growing use of thin film cells in the PV industry. It must be



# Photovoltaics

## Approximate Amounts (2010)

	Thickness (microns), g/m <sup>2</sup> (90% use)	g/W	Cost (\$/g, \$/W, \$/m <sup>2</sup> ; bulk, pure)	Competing Uses	Concern (mid- & long-term)
<b>Te</b>	3 um; 10-12 g	0.1 g/W (@ 110 W/m <sup>2</sup> )	\$0.3/g, \$0.03/W, \$3.3/m <sup>2</sup>	Few and small	Price & Supply, and Intrinsic Availability
<b>In</b>	1-2 um; 2-4 g (In/Ga = 4)	0.02-0.04 g/W (@ 110 W/m <sup>2</sup> )	\$0.5/g, \$0.02/W, \$1.5/m <sup>2</sup>	LCD (large and valuable)	Price & Supply, and Intrinsic Availability
<b>Ga</b>	1-2 um; 0.5-1 g (In/Ga =4)	0.005-0.01 g/W (@ 110 W/m <sup>2</sup> )	\$0.5/g, \$0.005/W, \$0.4/m <sup>2</sup>	Few and small	Price & Supply, and Intrinsic Availability
<b>Mo</b>	1 um; 10 g	0.1 g/W (@ 110 W/m <sup>2</sup> )	\$0.02/g, \$0.002/W, \$0.2/m <sup>2</sup>	Large	Price
<b>Se</b>	1-2 um; 3.5-7 g/m <sup>2</sup>	0.03-0.06 g/W (@ 110 W/m <sup>2</sup> )	\$0.1/g, \$0.005/W, \$0.5/m <sup>2</sup>	Large and valuable	Price & Supply, and Intrinsic Availability
<b>Ag</b>	30 g/m <sup>2</sup>	0.2 g/W (@ 160 W/m <sup>2</sup> )	\$0.5/g, 0.09/W, \$15/m <sup>2</sup>	Large and valuable	Price
<b>Cu</b>				Large and valuable	Price

# Photovoltaics

## Approximate Future (2030) Amounts

	Thickness (microns), g/m <sup>2</sup> (90% use)	g/W	Cost (\$/g, \$/W, \$/m <sup>2</sup> ; bulk, pure feedstock), 2010 prices	Competing Uses	How?
<b>Te</b>	2/3 um; 2 g	0.013 g/W (@ 150 W/m <sup>2</sup> )	\$0.5/g, \$0.007/W, \$1/m <sup>2</sup>	Few and small	Thinner CdTe, Higher Efficiency
<b>In</b>	0.75 um; 1.5 g (In/Ga = 4)	0.01 g/W (@ 160 W/m <sup>2</sup> )	\$1/g, \$0.01/W, \$1.5/m <sup>2</sup>	LCD (large and valuable)	Thinner CuInSe <sub>2</sub> Alloy, Higher Efficiency
<b>Ga for CIS</b>	0.75 um; 0.4 g (In/Ga = 4)	0.0025 g/W (@ 160 W/m <sup>2</sup> )	\$1/g, \$0.0025/W, \$0.4/m <sup>2</sup>	Few and small	Thinner CuInSe <sub>2</sub> Alloy, Higher Efficiency
<b>Mo</b>	0.5 um; 5 g	0.03 g/W (@ 160 W/m <sup>2</sup> )	\$0.02/g, \$0.0006/W, \$0.1/m <sup>2</sup>	Large	Thinner Mo contact, Higher Efficiency
<b>Se</b>	0.75 um; 2.6 g/m <sup>2</sup>	0.016 g/W (@ 160 W/m <sup>2</sup> )	\$0.2/g, \$0.003/W, \$0.5/m <sup>2</sup>	Large and valuable	Thinner CuInSe <sub>2</sub> Alloy, Higher Efficiency
<b>Ag</b>	15 g/m <sup>2</sup>	0.07 g/W (@ 220 W/m <sup>2</sup> )	\$1/g, 0.1/W, \$15/m <sup>2</sup>	Large and valuable	Higher Efficiency
<b>Cu</b>				Large and valuable	

# Indium in solar panels

- 50 metric tons required for enough solar panels to provide 1 gigawatt of energy
- \$500/kg in 2009
- 2008—US used 800 megawatts of energy by solar panels connected to the grid (0.1% total US energy)
- 600,000 metric tons reserves in the world in 2009
  - Zinc sulfide deposits
  - Tin-tungsten veins
  - Porphyry copper deposits



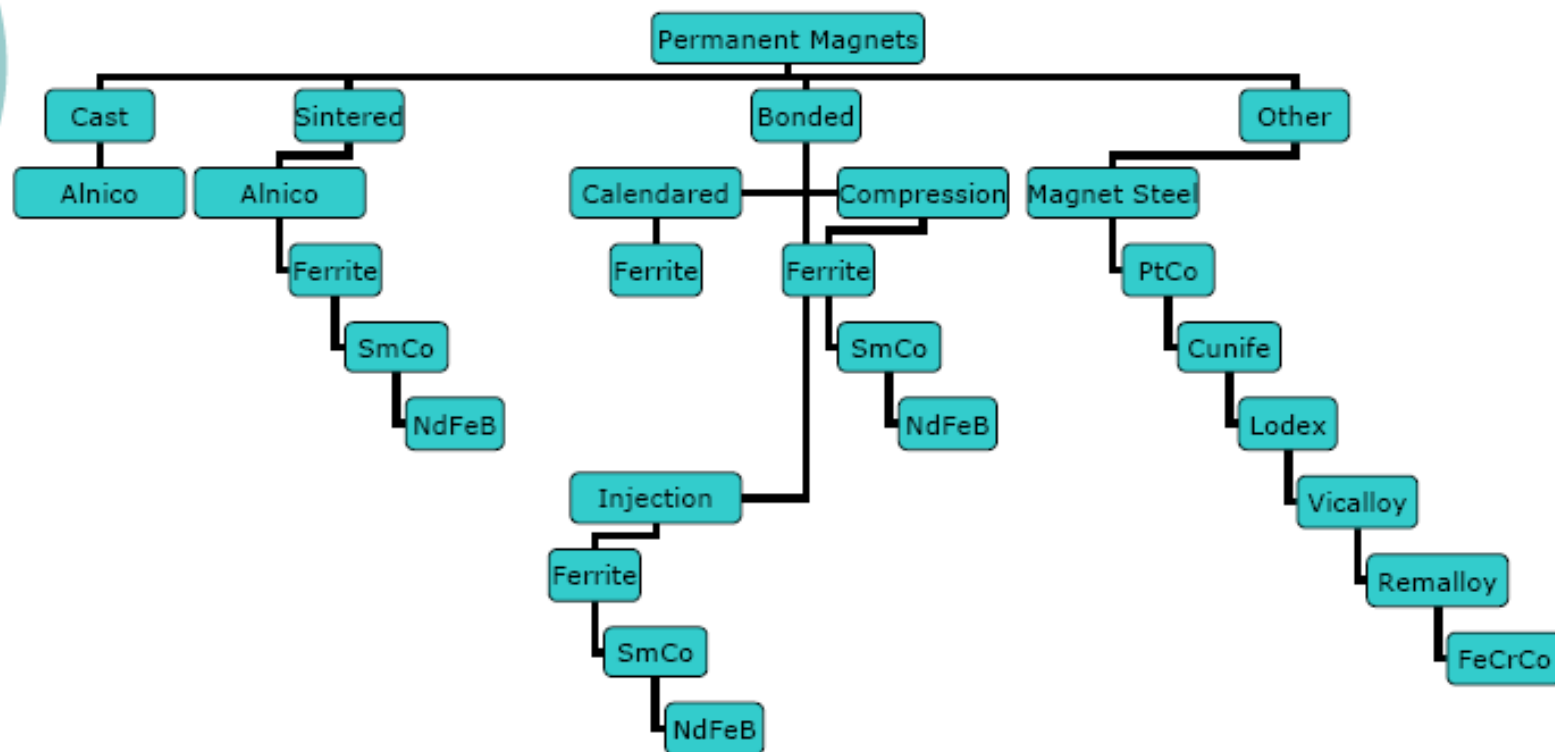
Ingots of indium.

# MAGNETS

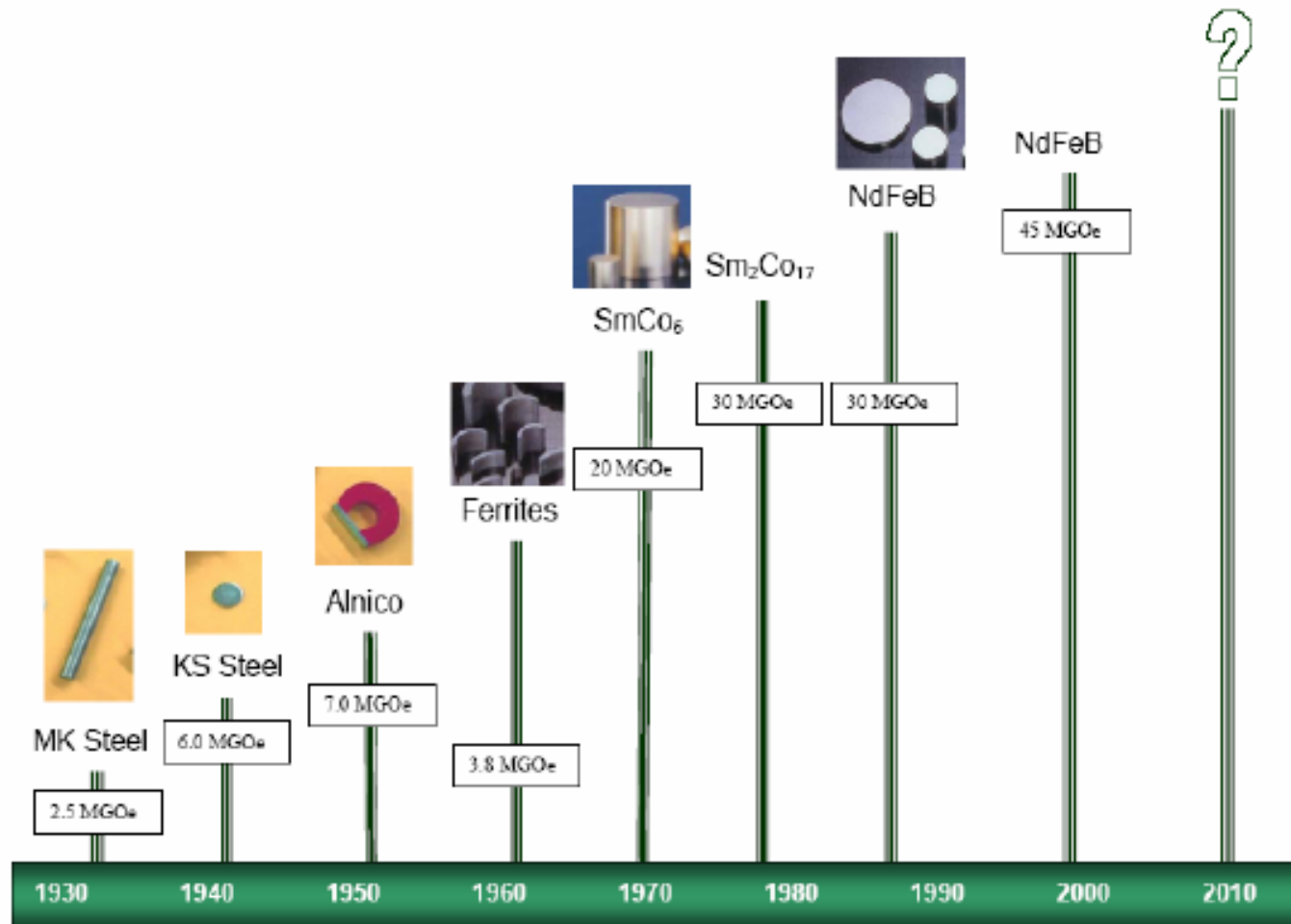
# Permanent Magnets

- Automotive
  - Electronics
  - Appliances
  - Medical
  - Military
  - Aerospace
  - Automation
  - Wind turbines
- Fe
  - Sm-Co
  - Nd-Fe-B
  - Cu-Ni-Fe
  - Fe-Cr-Co
  - Al-Ni-Co

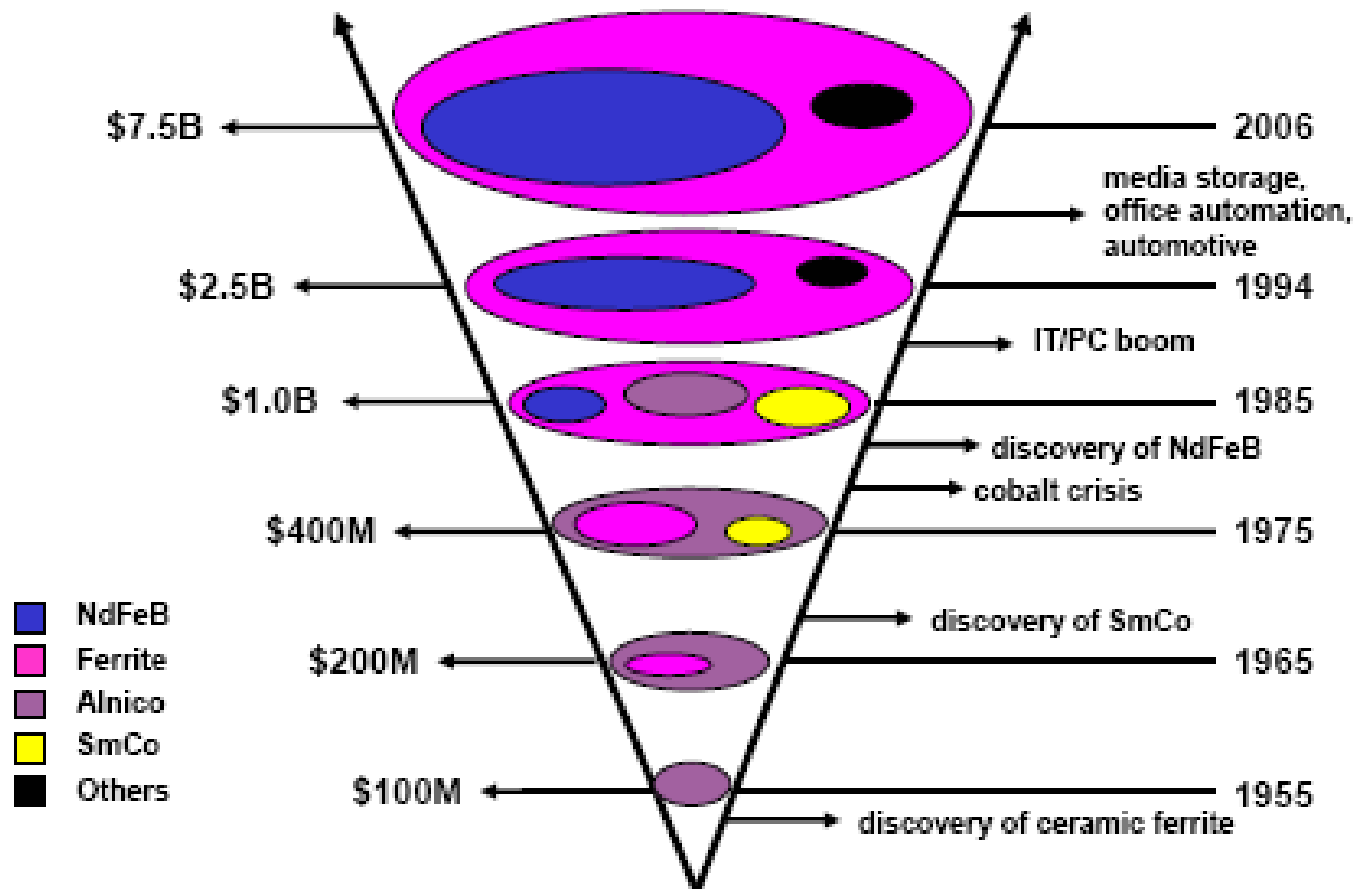
# The Magnet Family



# History of Magnetic Materials



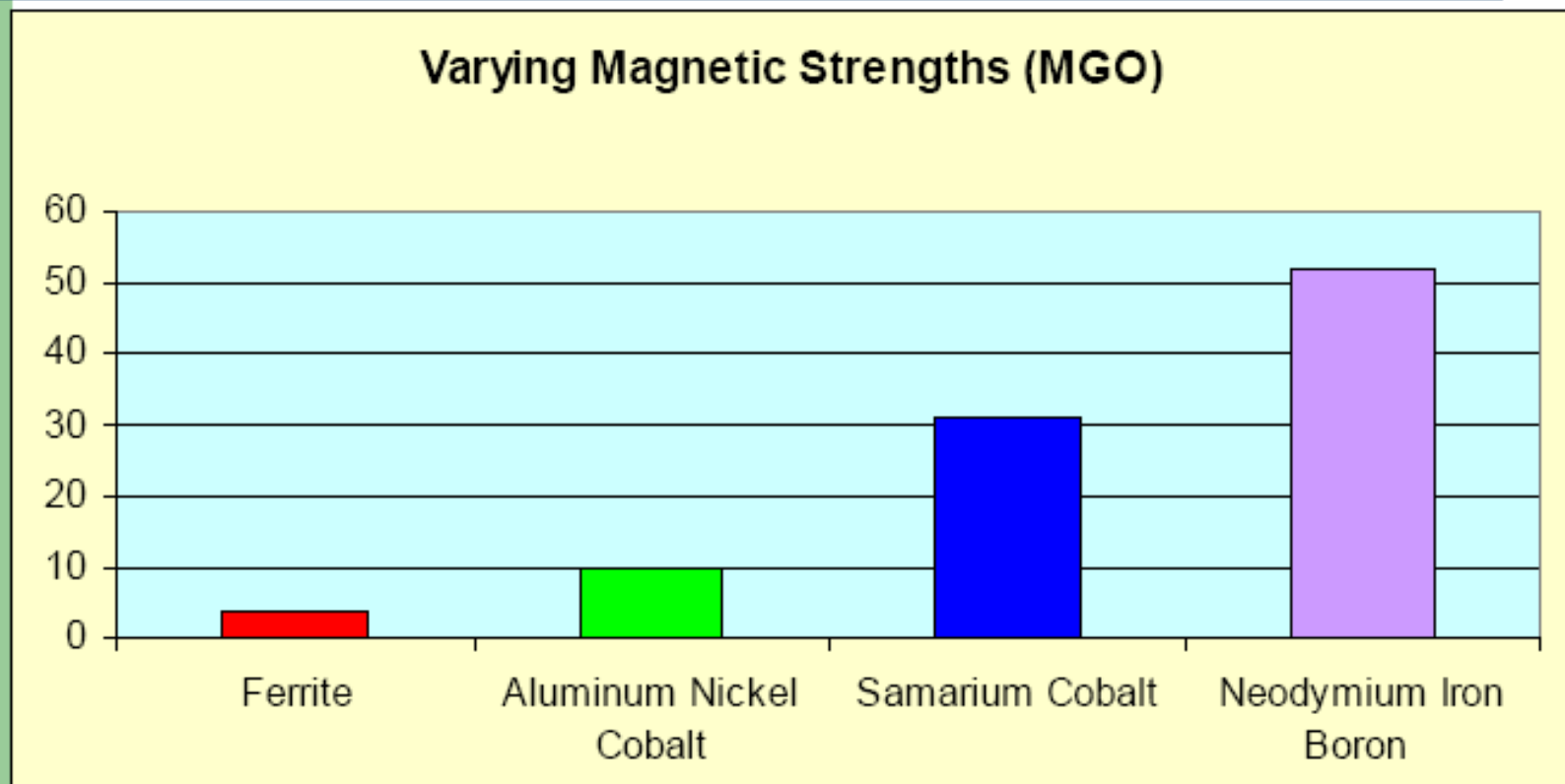
## Global value growth of permanent magnets



2007 Magnetism Conference



# Permanent Magnet Types & Strength



Rare earths

# WIND TURBINES

## FACT SHEET

## WIND TURBINES

### What Mineral Products & Metals Are Needed To Make Wind Turbines?



- **Aggregates and Crushed Stone (for concrete):** Mined in the United States.
- **Bauxite (aluminum):** Mined in Australia, China, Brazil, India, Guinea, Jamaica, Russia, Venezuela, Suriname, Kazakhstan, Guyana and Greece.
- **Clay and Shale (for cement):** Mined in United States.
- **Coal (by-product coke is used to make steel):** Coal is mined world-wide, and constitutes 45% of the generation of U.S. electricity.
- **Cobalt (magnets):** Mined in Congo-Kinshasa, Canada, Zambia, Russia, Australia, China, Cuba, Morocco, New Caledonia and Brazil.
- **Copper (wiring):** Mined in Chile, United States, Peru, China, Australia, Russia, Indonesia, Canada, Zambia, Poland and Mexico.
- **Gypsum (for cement):** Mined in China, United States, Iran, Spain, Thailand, Japan, Canada, Italy, France, Australia, Turkey, India, Russia, Saudi Arabia, Brazil, Egypt, Germany, United Kingdom, Algeria, Poland, Argentina and Austria.
- **Iron ore (steel):** Mined in China, Brazil, Australia, India, Russia, Ukraine, United States, South Africa, Iran, Canada, Sweden, Kazakhstan, Venezuela and Mexico.
- **Limestone (for cement):** Mined in United States.
- **Molybdenum (alloy in steel):** Mined in China, United States, Chile, Peru, Mexico, Canada, Armenia, Iran, Russia and Mongolia.
- **Rare Earth Oxides (magnets; batteries):** Mined in China, India and Brazil.

- **Zinc (galvanizing):** Mined in China, Peru, Australia, United States, Canada, India, Kazakhstan, Ireland and Mexico.
- **Silica Sand (for cement):** Mined in United States, Italy, Germany, United Kingdom, Australia, France, Spain, Japan, Poland, Hungary, South Africa, Mexico, Austria, Iran, Republic of Korea, Slovakia, Canada, Belgium, India, Bulgaria, Norway, Chile, Gambia, Turkey and Czech Republic.

### INTERESTING FACTS

- The foundation may contain over a thousand tons of concrete and rebar. Dimensions are often between 30-50 ft. across and 6-30 ft. deep.
- The average tower height is 229 ft. 8 in. Shafts are sometimes driven down farther to help anchor it. The platform is critical to stabilizing the immense weight of the turbine assembly.
- Depending on the model, industrial wind turbines can weigh between 164-334 tons, or more.
- 5,000 commercial-scale wind turbines were installed in 2008.
- Like old fashioned windmills, today's wind turbines use blades to collect the kinetic energy of the wind. The wind flows over the blades creating lift, like the effect on airplane wings, which causes them to turn. The blades are connected to a drive shaft that turns an electric generator to produce electricity.
- When the wind isn't blowing, other types of power plants must be used to make electricity.
- Wind turbines also use neodymium, boron and iron magnets in their construction and operation.
- The U.S. possesses the largest non-China rare earth resource in the world at the Mt. Pass Mine in California.



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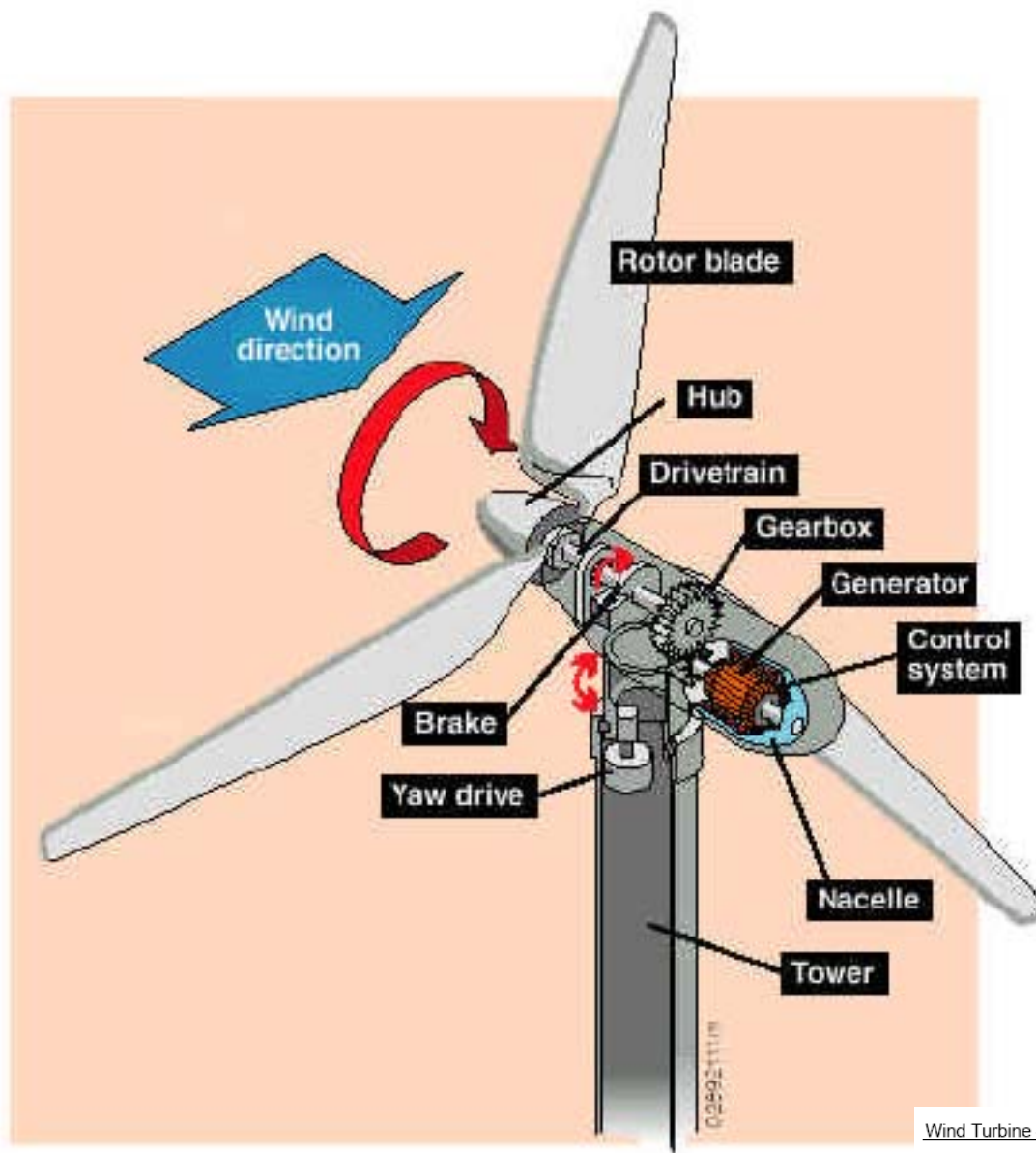
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# Wind turbines



**Wind Turbine Nomenclature**

Wind Turbine – Materials and Manufacturing Fact Sheet

August 29, 2001

## **Wind Turbine - Materials and Manufacturing Fact Sheet**

Prepared for the Office of Industrial Technologies, US Department of Energy  
By Princeton Energy Resources International, LLC.  
Dan Ancona and Jim McVeigh

**Table 1. Turbine Component Weight and Cost**

<b>Component</b>	<b>% of Machine Weight</b>	<b>% of Machine Cost [5]</b>
Rotor	10-14	20-30
Nacelle and machinery, less	25-40	25
Gearbox and drivetrain	5-15	10-15
Generator systems	2-6	5-15
Weight on Top of Tower	35-50	N/A
Tower	30-65	10-25

**Table 3. Percentage of Materials Used in Current Wind Turbine Component**

Component/ Material (% by weight)	Large Turbines and ( <i>Small Turbines</i> <sup>1</sup> )							Carbon Filament Reinforced Plastic <sup>4</sup>
	Permanent Magnetic Materials	Pre- stressed Concrete	Steel	Aluminum	Copper	Glass Reinforced Plastic <sup>4</sup>	Wood Epoxy <sup>4</sup>	
<b>Rotor</b>								
Hub			(95) - 100	(5)				
Blades			5			95	(95)	(95)
<b>Nacelle</b> <sup>2</sup>	(17)		(65) - 80	3 - 4	14	1 - (2)		
Gearbox <sup>3</sup>			98 - (100)	(0) - 2	(<1) - 2			
Generator	(50)		(20) - 65		(30) - 35			
<b>Frame, Machinery &amp; Shell</b>			85 - (74)	9 - (50)	4 - (12)	3 - (5)		
<b>Tower</b>		2	98	(2)				

Notes:

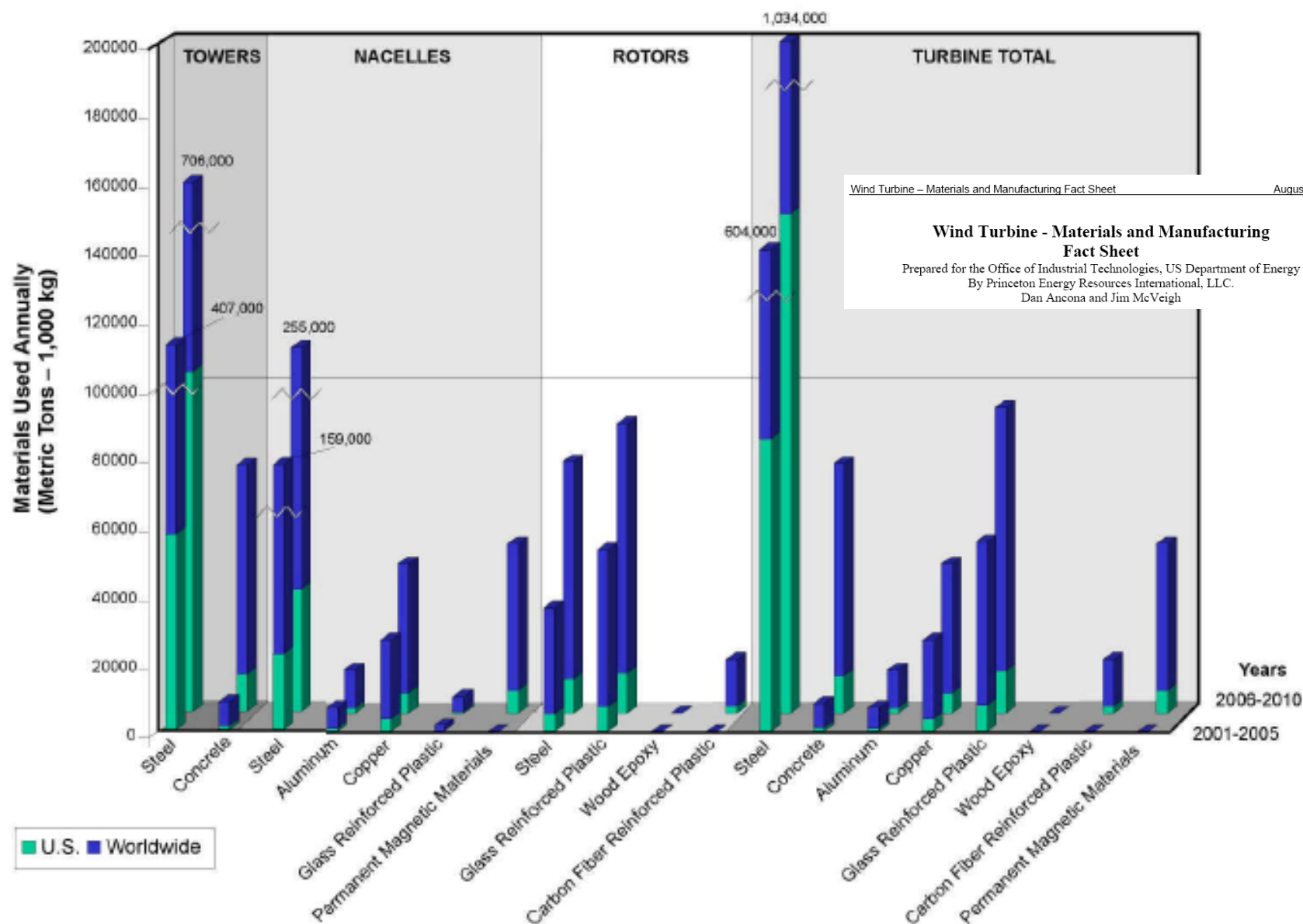
1. Small turbines with rated power less than 100 kW- (listed in italics where different)
2. Assumes nacelle is 1/3 gearbox, 1/3 generator and 1/3 frame & machinery
3. Approximately half of the small turbine market (measured in MW) is direct drive with no gearbox
4. Rotor blades are either glass reinforced plastic, wood-epoxy or injection molded plastic with carbon fibers

## Wind Turbine - Materials and Manufacturing Fact Sheet

Prepared for the Office of Industrial Technologies, US Department of Energy  
By Princeton Energy Resources International, LLC.  
Dan Ancona and Jim McVeigh

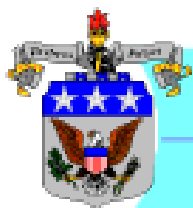


# Wind Turbine Materials Usage





# ENGINES



# *Minerals in Pratt & Whitney F100 Turbofan Engine*

Center for  
Strategic  
Leadership

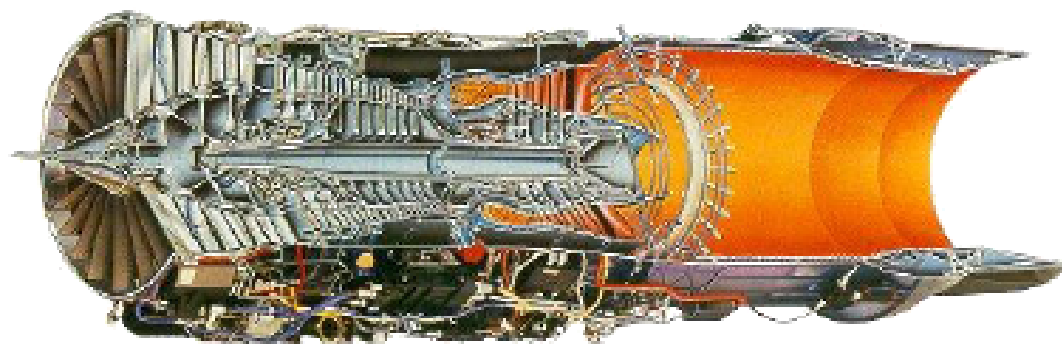
# CSL

Nickel  
4,504 Lbs.

Titanium  
5,440 Lbs.

Chromium  
1,485 Lbs

F100-PW-229 TURBOFAN ENGINE



Columbium  
145 Lbs.

Manganese  
23 Lbs.

Cobalt  
885 Lbs

# CHALLENGES

# GENERAL COMMENTS

- Many of these minerals do not require the tonnages we are used to mine for metals like Fe, Cu, Pb, Zn—i.e. smaller deposits

Commodity	US production 2009 mt	World production 2009 mt	consumption 2009 mt	Price 2009	World reserves 2009 mt
Cu	1,190,000	15,800,000	1,660,000	\$2.3/lb	540,000,000
Au	210	2,350	170	\$950/oz	47,000
REO	0	124,000	7,410	varies	99,000,000
Be	120	140	140	\$120/lb	15900+
Sb	0	187,000	22,400	\$2.3/lb	2,100,000
As	385	52,500	3,600	\$0.92/lb	1,070,000
Bi	100	7,300	1,020	\$7.4/lb	320,000
Ga	0	78	20	\$480/kg	1,000,000
Ge	5	14	5	\$950/kg	450+
Te	W		W	\$145/kg	22,000
cement	71,800,000	2,800,000,000	73,800,000	\$100/mton	

# The Promise of Clean Energy

## Applications for REPM

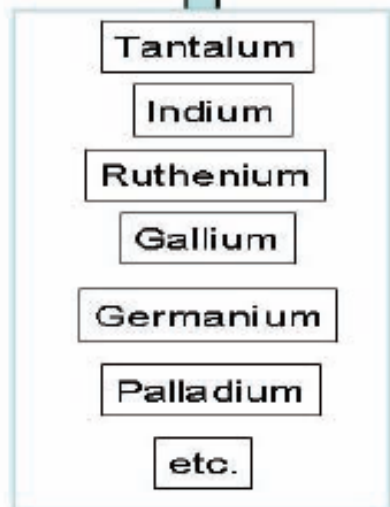
Applications		Typical Quantity of REO per unit
Traditional – disc drives, personal electronic devices, etc.		Grams
Hybrid and electric vehicles – direct drives and electric assist motors	 Chevy Volt	Kilograms
Direct Drive Wind Turbines		Metric Ton

<http://www.reitausa.org/storage/Challenges%20Facing%20New%20Global%20Rare%20Earth%20Separation%20Plants.pdf>

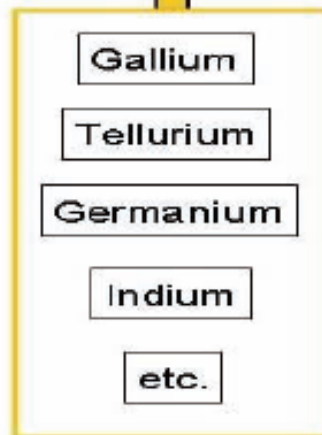
# GENERAL COMMENTS

- Some of these minerals are economically found in only 1-3 deposits in the world
- Some of these minerals are found in areas of the world that may not be economically unstable or particularly friendly to the U.S.
  - Minerals that provide major revenue to armed fractions for violence, such as that occurring in the Democratic Republic of Congo (GSA, Nov. 2010)
- Some of these minerals come only from the refining of metal deposits and are dependent upon that production
  - Many Cu and Au deposits utilize heap leach technology, which leaves other potential minerals unrecovered in the heap leach

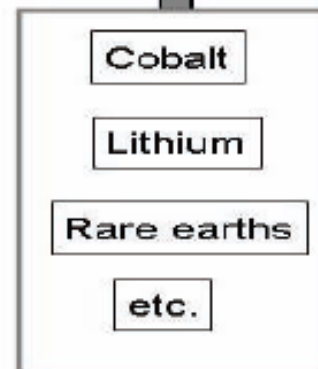
**EEE** (Electrical and electronics equipment)



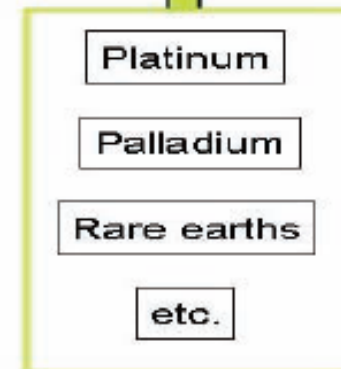
**photovoltaic**



**batteries**



**catalysts**



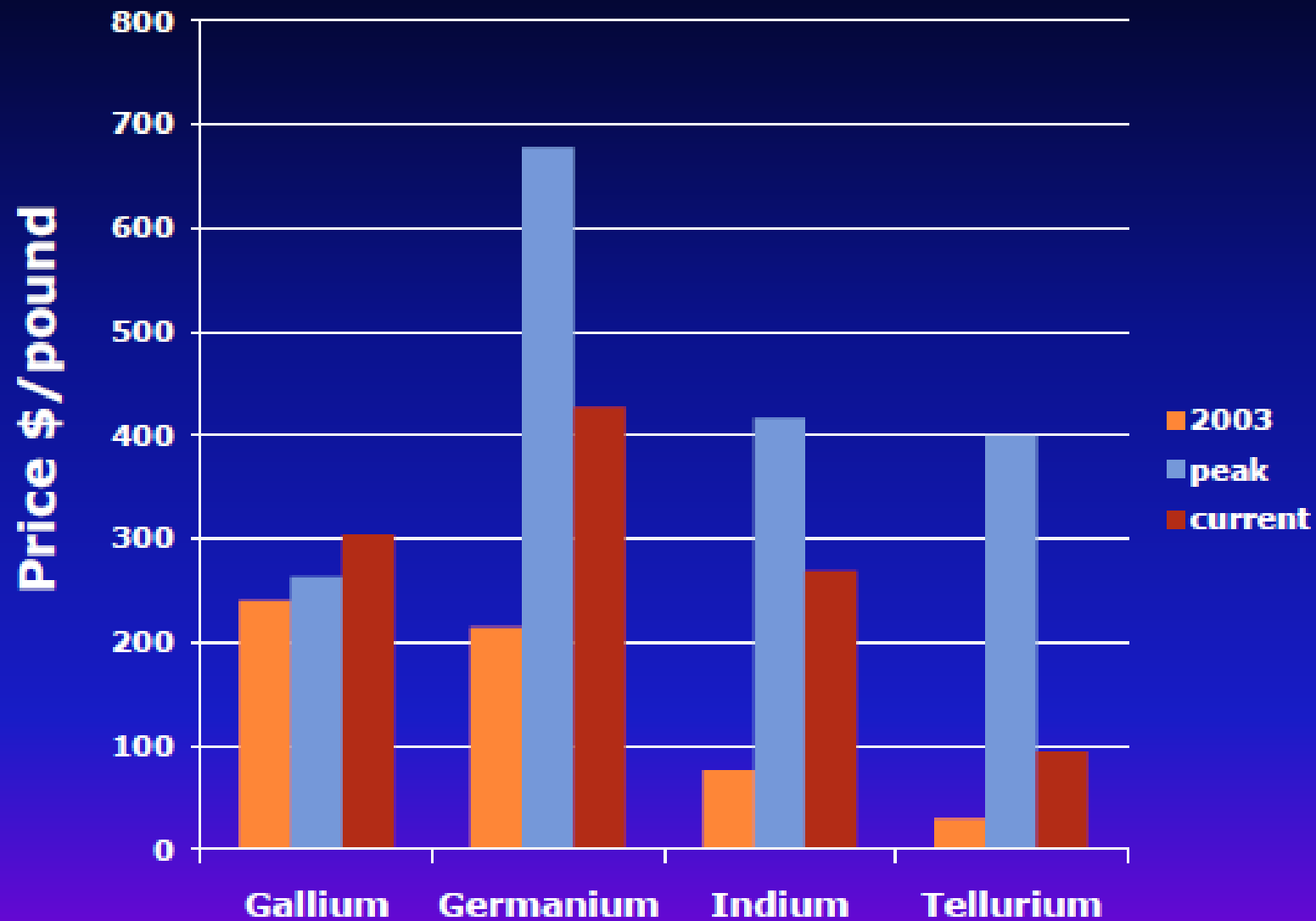
# Competing industries/uses



## **Key Issues with Critical Metals**

- Supply Concentration due to deposit nature and location
- Supply Concentration due to government subsidies
- Cartel style supply management – quotas and stockpiles
- Resource nationalism – domestic downstream beneficiation
- Long term cost volatility & price uncertainty
- Complex recovery processes – chemical plants
- State controlled capital investment
- Small size of market for some metals & the dread of over-supply
- End product Innovation as demand driver
- Policy as disruptive demand driver
- Efficiency innovation & Jevon's Paradox
- Substitution as demand destroyer
- Failure of just-in-time procurement strategy
- Opportunity Cost: geopolitical, domestic, end-user
- The Upstream Solution
- The Downstream Solution

# Price History



- Analytical labs are swamped (i.e., too long) and expensive
- There is a need for relatively quick, inexpensive methods to delineate drill hole targets
- Developing a procedure using a portable X-ray fluorescent instrument to use in drilling, stream sediment and soil surveys, to aid in exploration and ore control



## 5 Dimensions of Mineral Availability

### ***WHAT** Questions Must We Ask?*

1

**Geologic**  
Availability



✓ Does the mineral resource exist?

2

**Technical**  
Availability



✓ Can we extract and process it?

3

**Environmental  
& Social**  
Availability



✓ Can we produce it in environmentally and socially responsible and acceptable ways?

4

**Political**  
Availability



✓ How do governments influence availability through their policy and actions?

5

**Economic**  
Availability



✓ Can we produce it at a cost users are willing and able to pay?

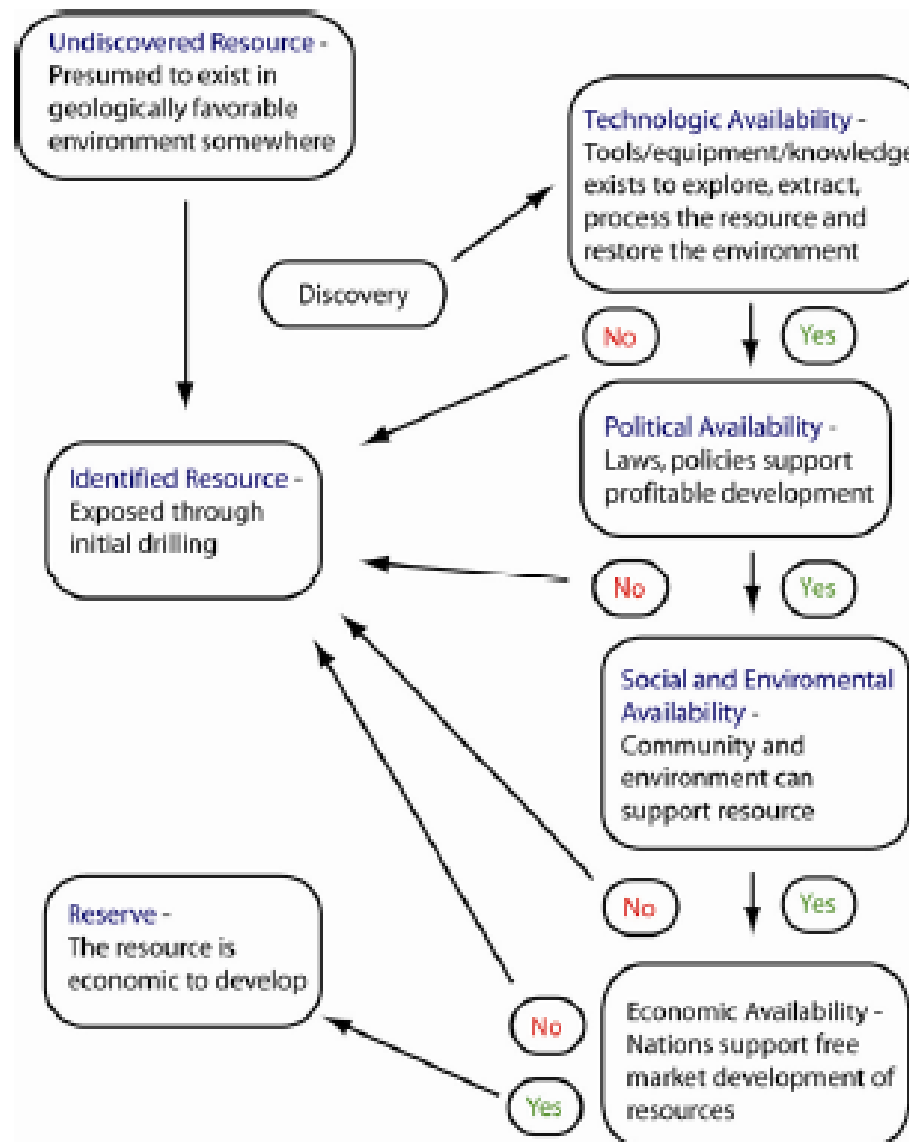
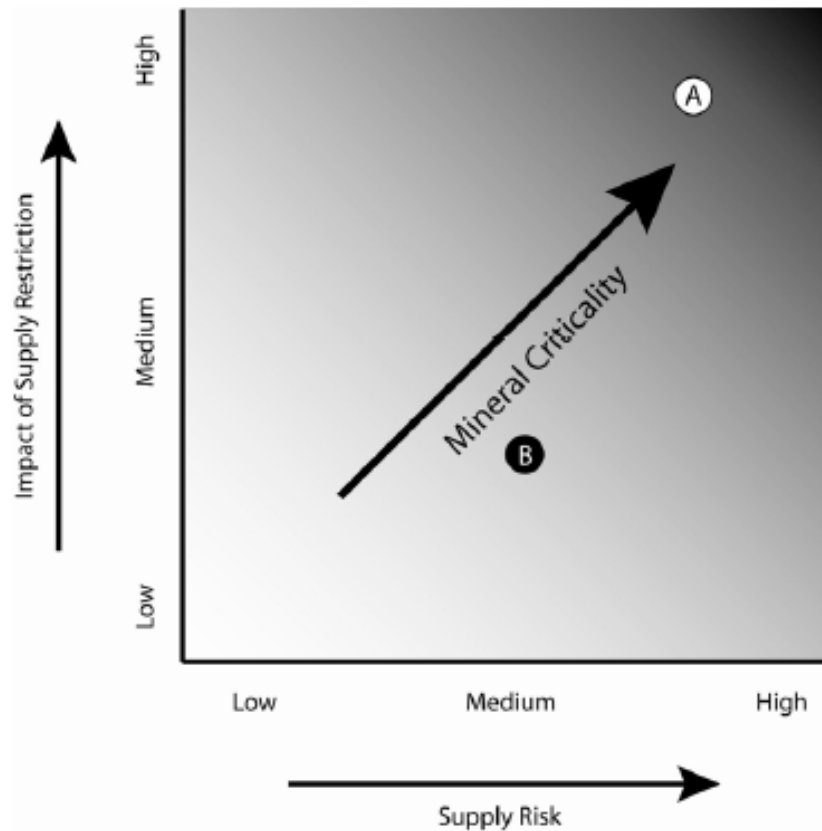


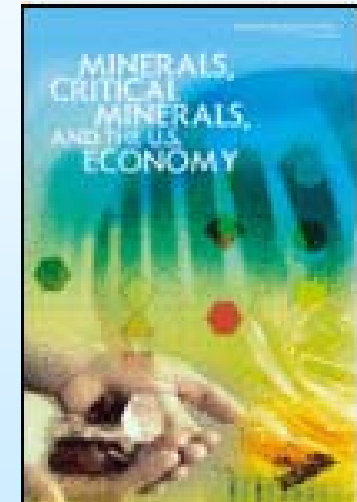
FIGURE 3.6. The availability of a mineral resource is dynamic in the five dimensions of geologic, technologic, social and environmental, political, and economic availability. Only if the extraction and processing of the resource is proved to be economically profitable is it considered a reserve.

[http://www.nma.org/pdf/101606\\_nrc\\_study.pdf](http://www.nma.org/pdf/101606_nrc_study.pdf)

# Criticality Index



**FIGURE 1.3** The criticality matrix as established in this report allows evaluation of the “criticality” of a given mineral. A specific mineral or mineral product can be placed on this figure after assessing the impact of the mineral’s supply restriction should it occur (vertical axis) and the likelihood of a supply restriction (horizontal axis). The degree of criticality increases as one moves from the lower-left to the upper-right corner of the figure. In this example, Mineral A is more critical than Mineral B. More specific descriptions of the parameters used to evaluate mineral supply restrictions and their impacts are presented in Chapters 2-4.



## MINERALS, CRITICAL MINERALS, AND THE U.S. ECONOMY

*Prepublication Version*

NATIONAL RESEARCH COUNCIL  
OF THE NATIONAL ACADEMIES

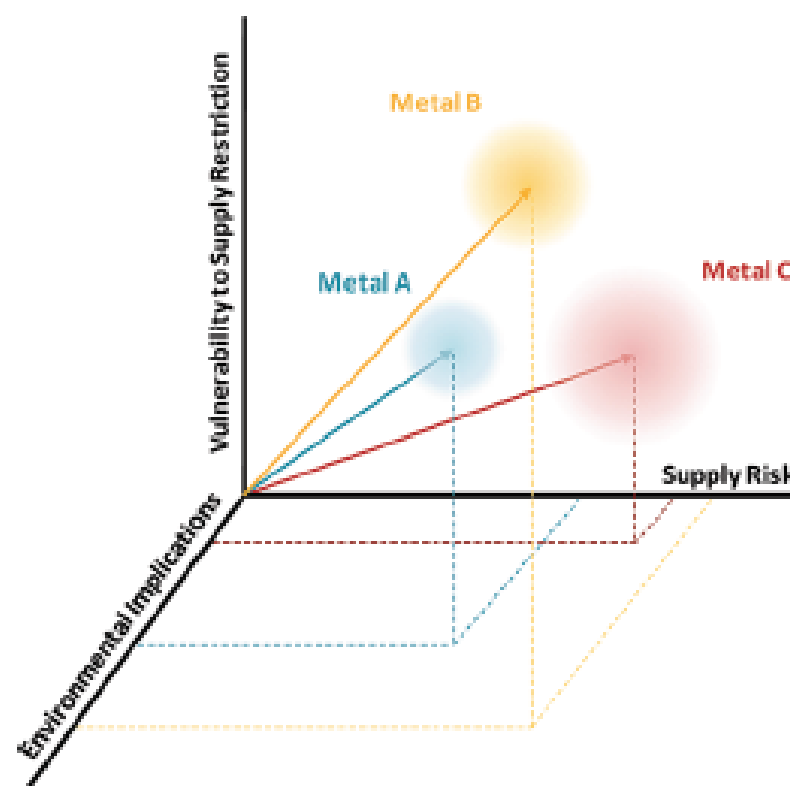
THIS PREPUBLICATION VERSION OF MINERALS, CRITICAL MINERALS, AND THE U.S. ECONOMY has been provided to the public to facilitate timely access to the committee’s findings. Although the substance of the report is final, editorial changes may be made throughout the text, and citations will be checked prior to publication. The final report will be available through the National Academies Press in the December/January timeframe.

[http://www.nma.org/pdf/101606\\_nrc\\_study.pdf](http://www.nma.org/pdf/101606_nrc_study.pdf)

# Criticality is context specific:

- ◆ What is critical for a given manufacturer or product may not be critical for another, what is critical for a state may not be critical for a country, and what is critical for national defense may be different than what is necessary to make a television brighter or less expensive.
- ◆ Recent studies have expanded the scope of criticality to include environmental and technological factors.

Graedel, T. E.; Barr, R.; Chandler, C.; Chase, T.; Choi, J.; Christoffersen, L.; Friedlander, E.; Henly, C.; Nassar, N. T.; Schechner, D.; Warren, S.; Yang, M.; Zhu, C., 2012, Methodology of metal criticality determination: Environ. Sci. Technol., 46, 1063• 1070.



**Table 1. Relevant Material-Related Characteristics for Different Organizational Levels**

		using corporation	using nation	global
1	focus	relevance to that firm's product line	relevance to national industry and population	all uses of a material, wherever they happen
2	time scale	1–5 years	5–10 years	10–100 years
3	supply potential	crucial	very important	very important
4	technological change	very important	worth consideration	impossible to predict
5	geopolitical factors	crucial	important	unimportant
6	social factors	moderately important	very important	unimportant
7	environmental implications	important	important	moderately important
8	intensity of competition	crucial	depends on national industry composition	unimportant

Graedel, T. E.; Barr, R.; Chandler, C.; Chase, T.; Choi, J.; Christoffersen, L.; Friedlander, E.; Henly, C.; Nassar, N. T.; Schechner, D.; Warren, S.; Yang, M.; Zhu, C., 2012, Methodology of metal criticality determination: Environ. Sci. Technol., 46, 1063–1070.





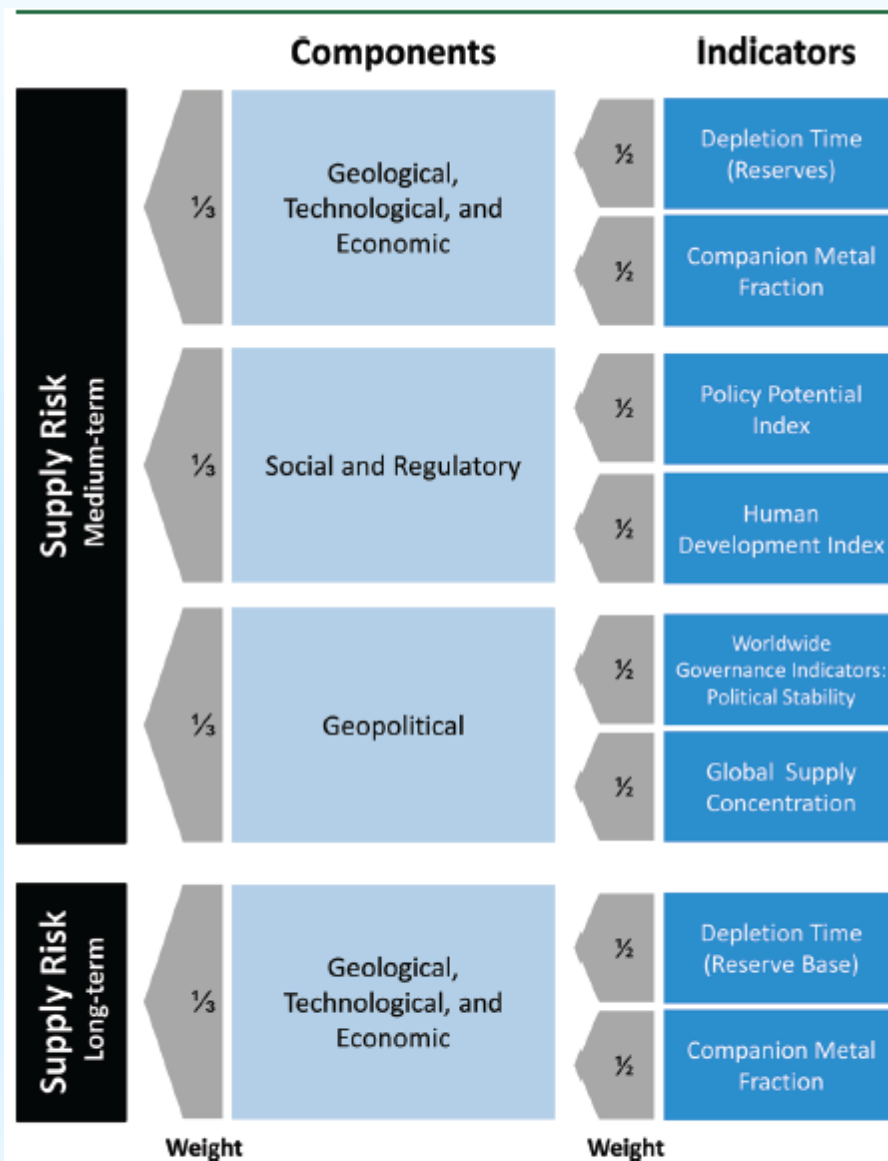
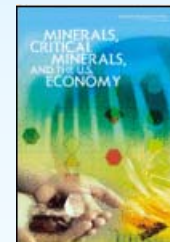
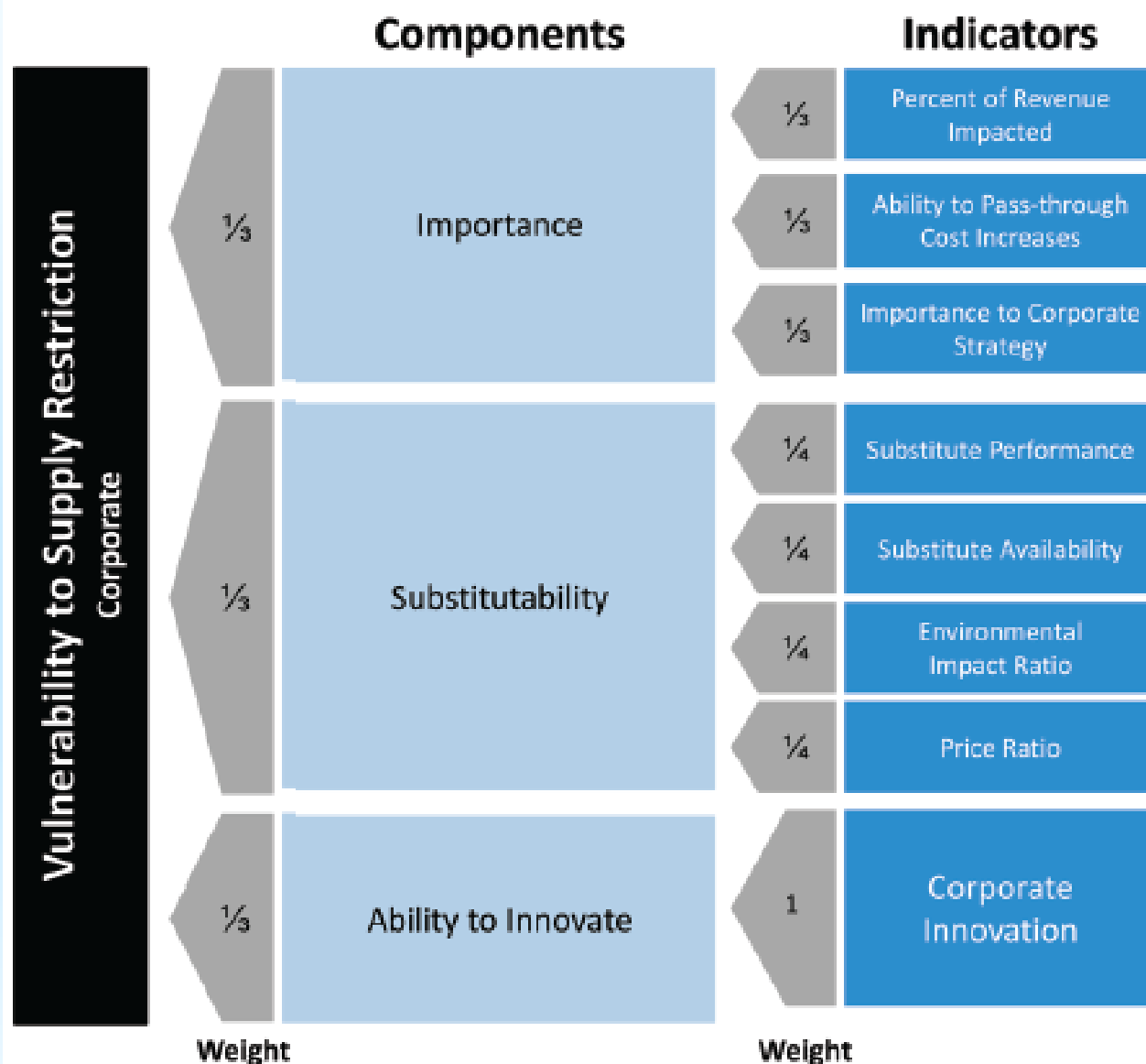
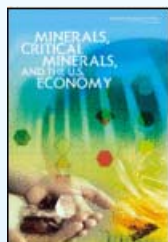


Figure 1. Diagram of the supply risk axis, its components, and its constituent indicators, for the medium-term perspective, used mainly in conjunction with the corporate- and national-level assessments and, for the long-term perspective, used mainly in conjunction with the global-level assessment.

Graedel, T. E.; Barr, R.; Chandler, C.; Chase, T.; Choi, J.; Christoffersen, L.; Friedlander, E.; Henly, C.; Nassar, N. T.; Schechner, D.; Warren, S.; Yang, M.; Zhu, C., 2012, Methodology of metal criticality determination: Environ. Sci. Technol., 46, 1063-1070.





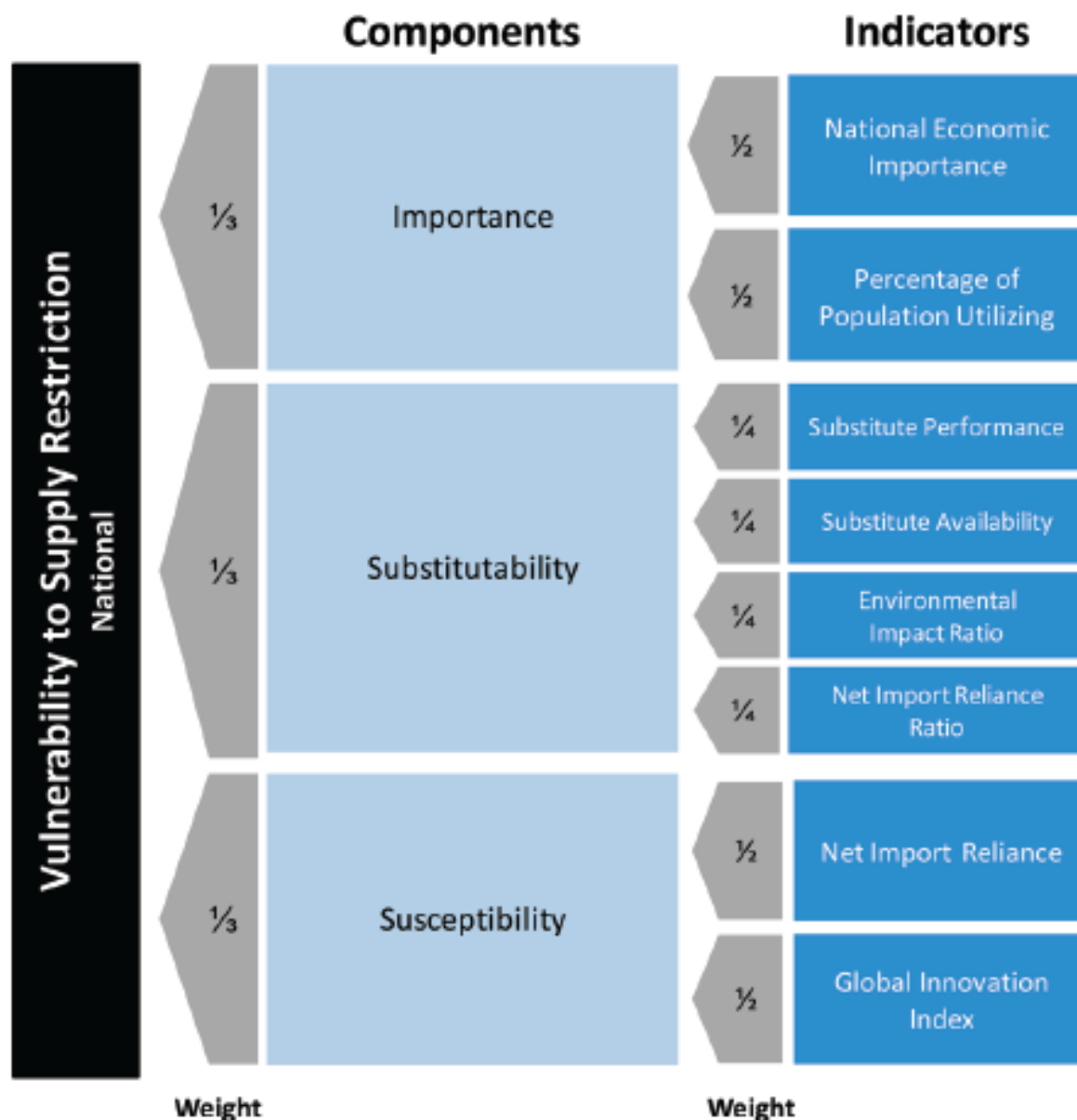
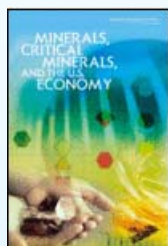
**Figure 2.** Components of the valuation methodology for the vulnerability to supply restriction, corporate level.

Table 2. Corporate-Level Vulnerability to Supply Restriction Matrix<sup>a</sup>

Component		Importance			Substitutability				Ability to Innovate
Indicator		Percentage of Revenue Impacted	Ability to Pass-through Cost Increases	Importance to Corporate Strategy	Substitute Performance	Substitute Availability	Environmental Impact Ratio	Price Ratio	Corporate Innovation
Score	87.5 (75-100)	> 5%	Practically impossible	Essential	Poor	Supply Risk score of substitute	See equation in SI	See equation in SI	Poor
	62.5 (50-75)	>2.5-5%	Difficult	Very important	Adequate				Adequate
	37.5 (25-50)	0.5-2.5%	Possible	Moderately important	Good				Good
	12.5 (0-25)	< 0.5%	Relatively easy	Less important	Exemplary				Exemplary

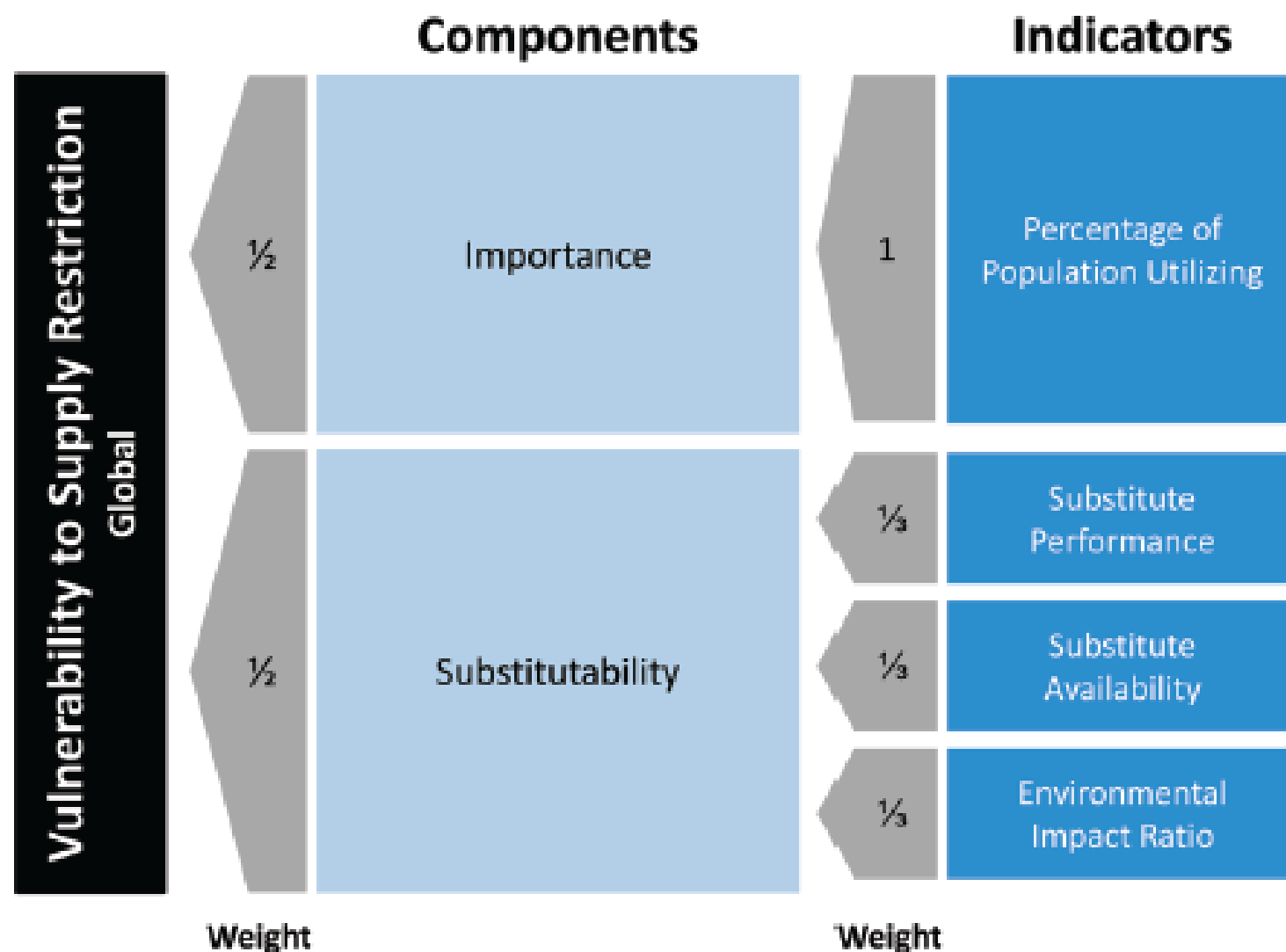
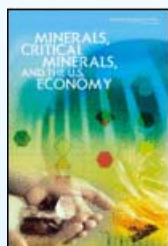
Graedel, T. E.; Barr, R.; Chandler, C.; Chase, T.; Choi, J.; Christoffersen, L.; Friedlander, E.; Henly, C.; Nassar, N. T.; Schechner, D.; Warren, S.; Yang, M.; Zhu, C., 2012, Methodology of metal criticality determination: Environ. Sci. Technol., 46, 1063-1070.





Graedel, T. E.; Barr, R.; Chandler, C.; Chase, T.; Choi, J.; Christoffersen, L.; Friedlander, E.; Henly, C.; Nassar, N. T.; Schechner, D.; Warren, S.; Yang, M.; Zhu, C., 2012, Methodology of metal criticality determination: Environ. Sci. Technol., 46, 1063-1070.

Figure 3. Components of the valuation methodology for the vulnerability to supply restriction, national level.



**Figure 4. Components of the valuation methodology for the vulnerability to supply restriction, global level.**

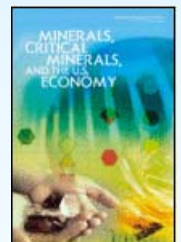
**TABLE 4.2** Scoring the Vertical Axis of the Criticality Matrix for Example Mineral X

Application Group (End Uses) for Mineral X	Proportion of Total U.S. Market for Mineral X in Application	Impact of Supply Restriction (Values of 1 to 4)	Weighted Score (Product of Columns 2 and 3)
Aerospace propulsion	0.27	4	1.08
Pigments	0.65	4	2.60
Biomedical devices	0.8	2	0.16
Overall importance in use	1.00 <sup>a</sup>	n.a.	3.84 <sup>b</sup>

<sup>a</sup> Total proportion will always equal 1.

<sup>b</sup> Final weighted score.

NOTE: n.a. = not applicable.



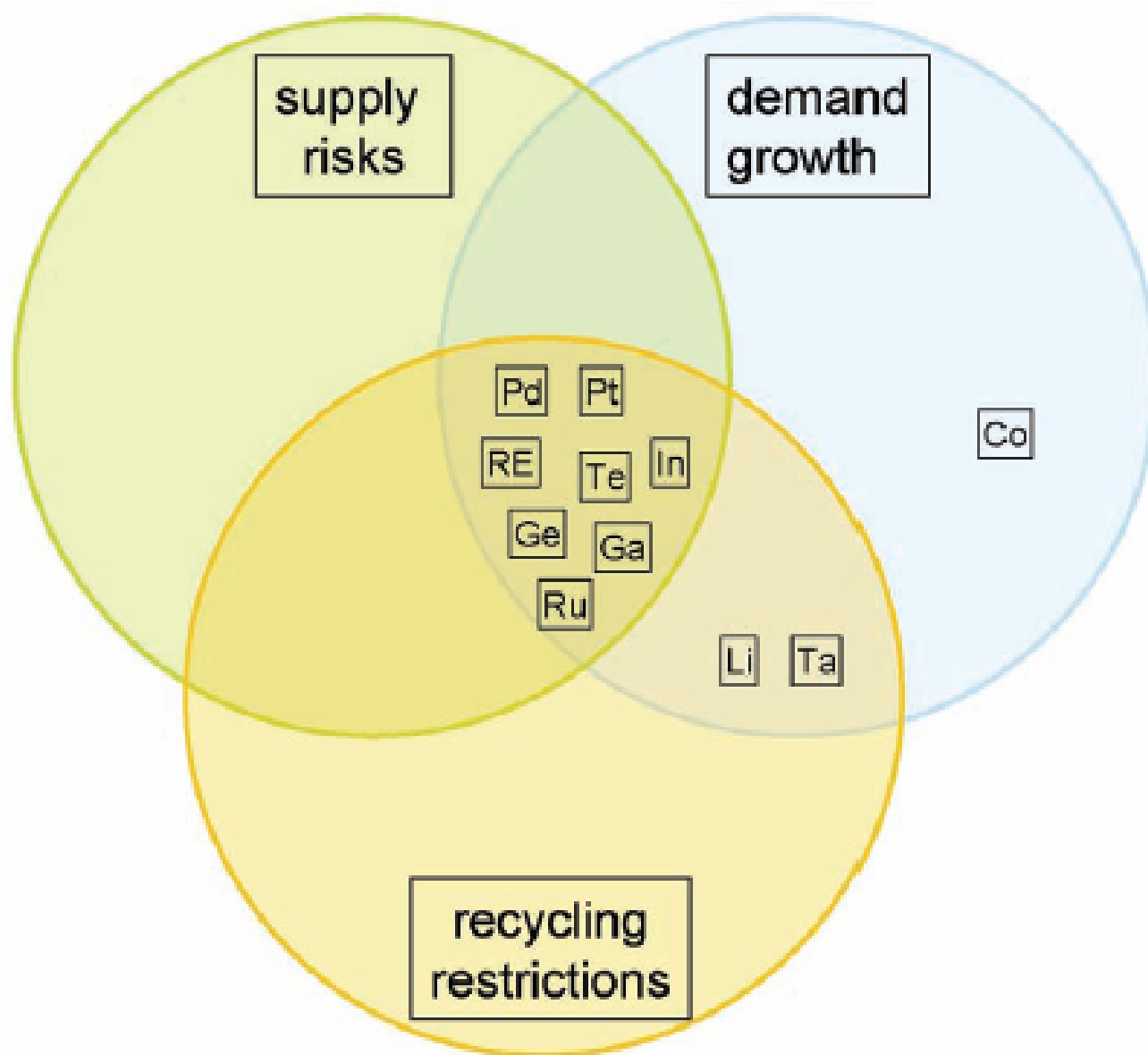


Figure 7.1: Prioritization step 1: general assignment of critical metals

	<b>Rapid demand growth</b> > 50% increase of total demand until 2020	<b>Moderate demand growth</b> > 20% increase of total demand until 2020
✓ Tantalum	+	
✓ Indium	+	
✓ Ruthenium	+	
✓ Palladium	+	
✓ Platinum	+	
✓ Rare earths	+	
✓ Gallium	+	
✓ Tellurium	+	
Germanium		+
Cobalt		+
✓ Lithium	+	

Figure 7.2 Prioritization step 2: Distinction of rapid/moderate demand growth



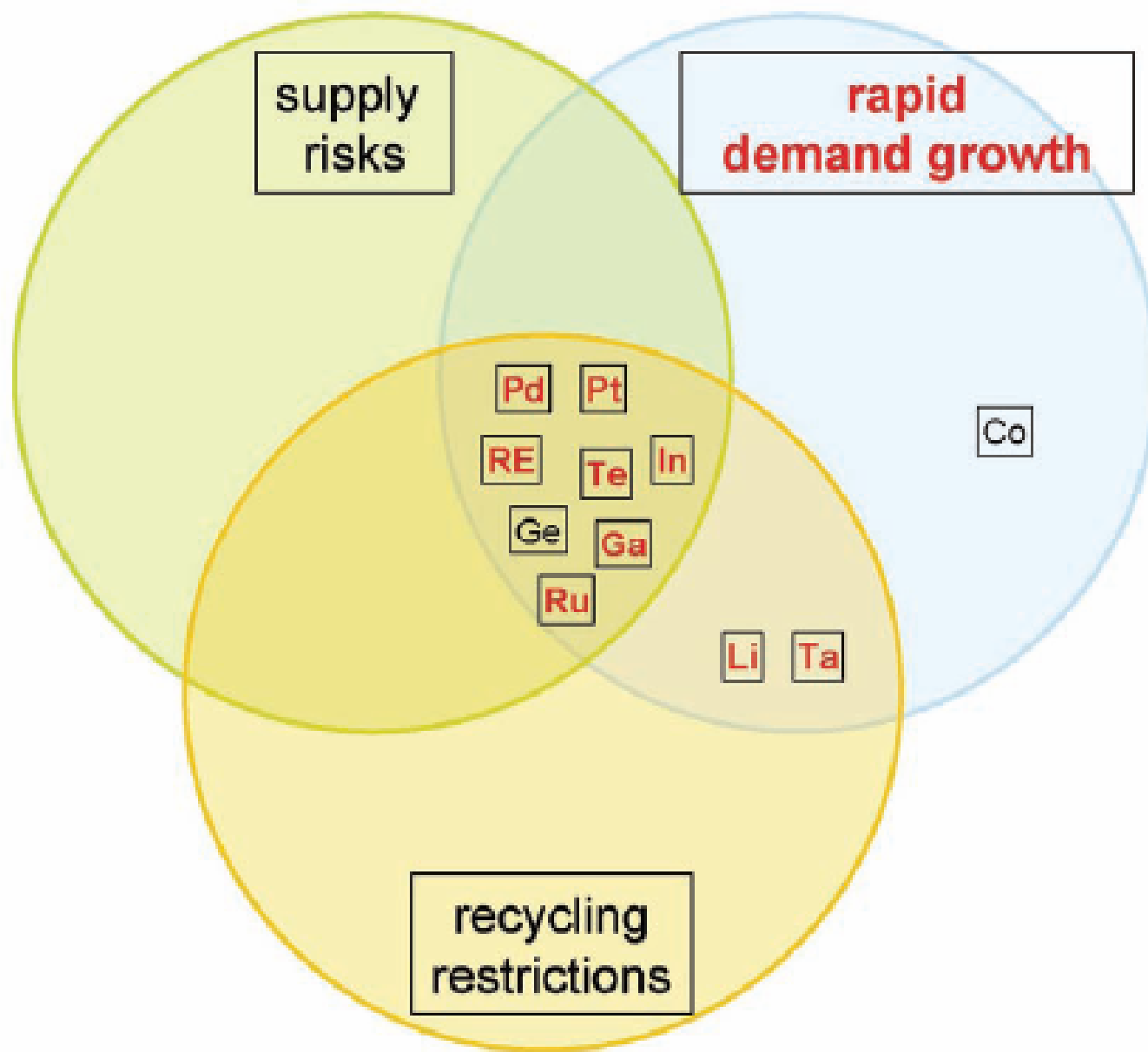


Figure 7.3: Visual transfer of differentiated growth rates for demand

	<b>Regional concentration of mining</b> >90% share of global mining in the 3 mayor countries	<b>Physical scarcities</b> reserves compared to annual demand	<b>Temporary scarcity</b> Time lag between production and demand	<b>Structural or technical scarcity</b> Metal just a minor or by-product
Tantalum	-	-	+-	-
Indium	-	+	+-	+
Ruthenium	+	-	+-	-
Palladium	+	-	+-	-
Platinum	+	-	+-	-
Rare earths	+	-	-	-
Gallium	-	-	+	+
Tellurium	-	+-	+	+
Germanium	-	+- (?)	+- (?)	+
Cobalt	-	-	-	+-
Lithium	-	-	+-	-

Figure 7.4: Prioritization step 3: Focused distinction of supply restriction

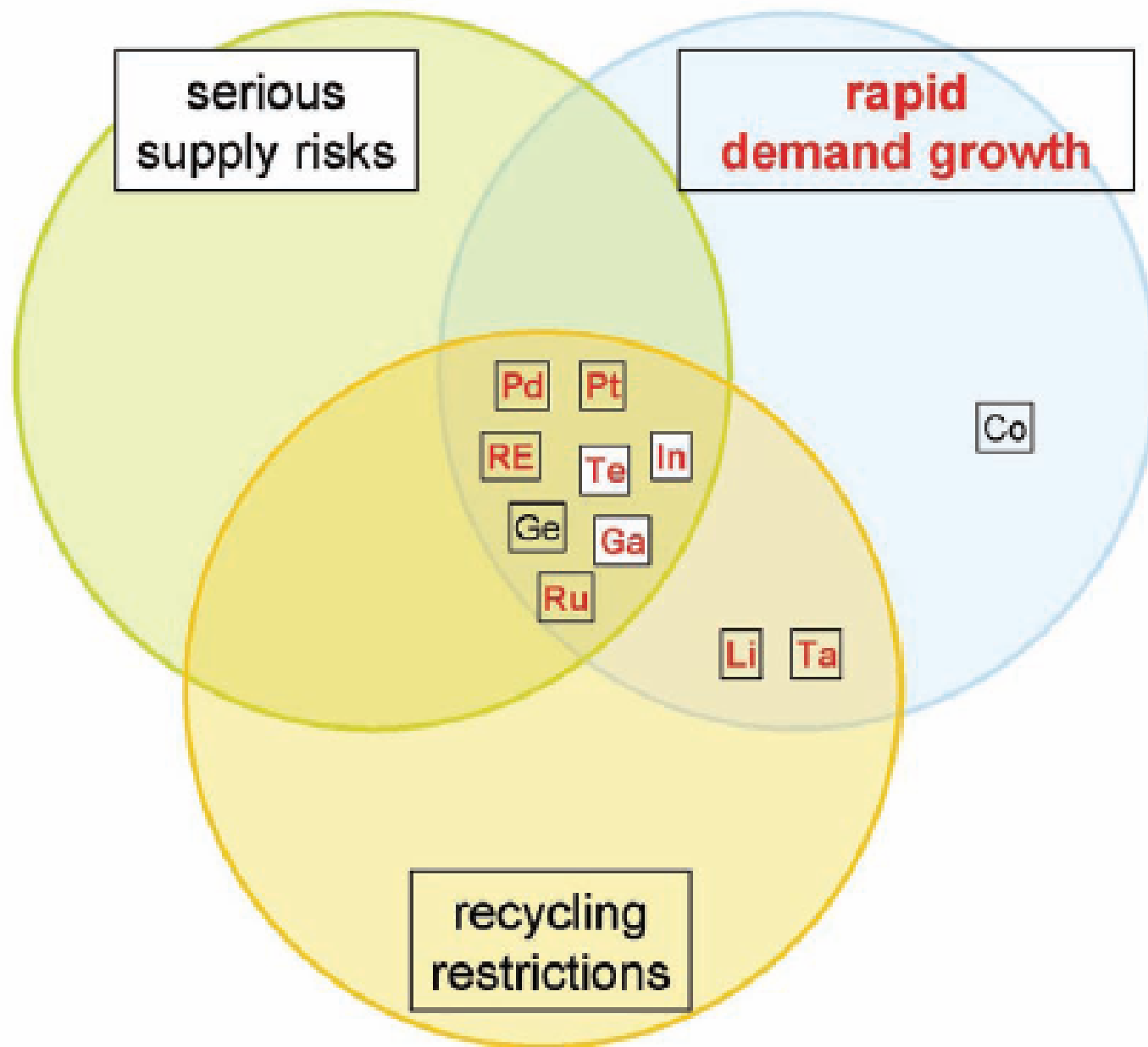


Figure 7.5: Visual transfer of differentiated supply restrictions

## 7.4 Focused prioritization regarding recycling (4<sup>th</sup> step)

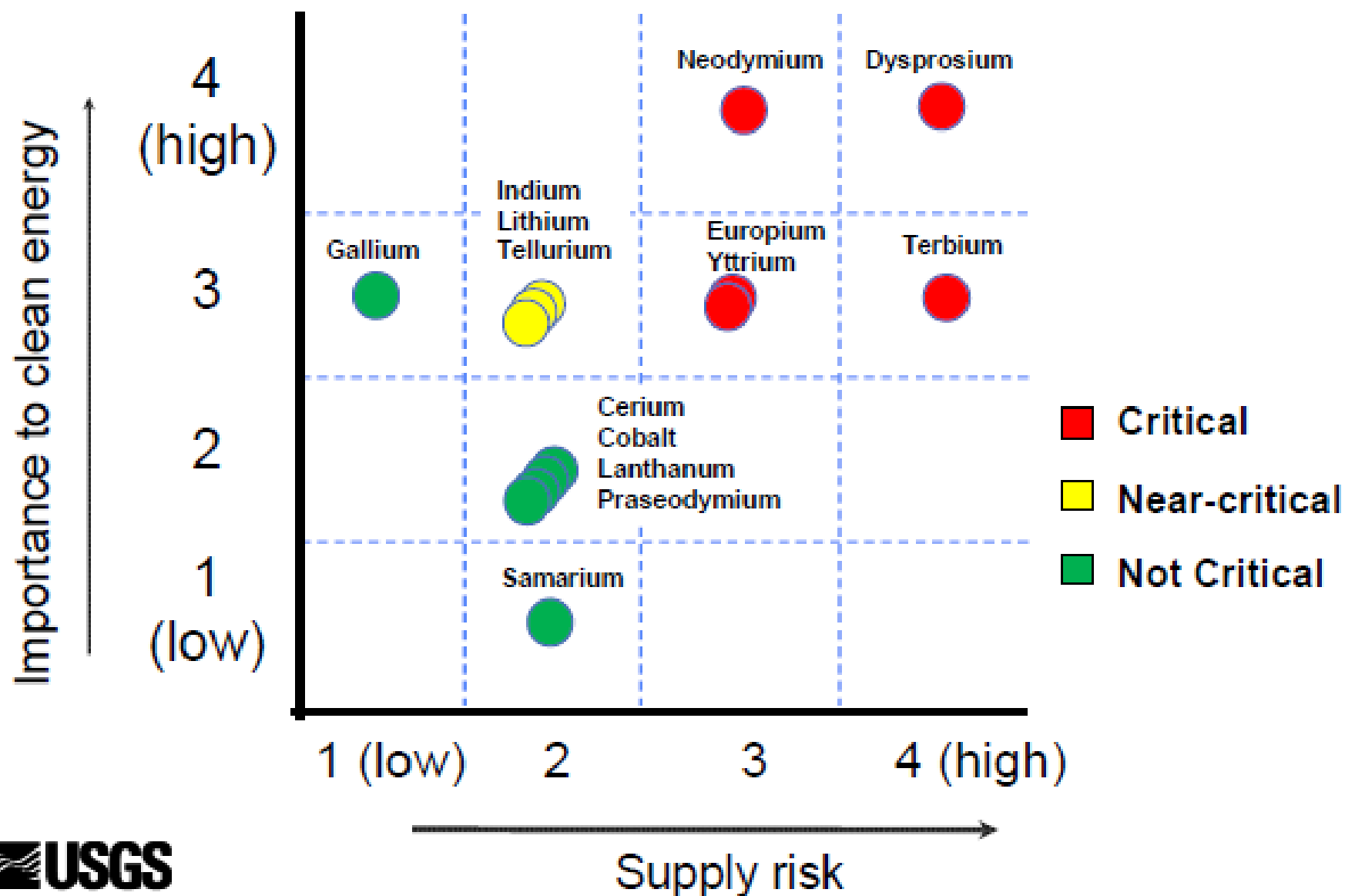
	High scale of dissipative applications	Physical/chemical limitations for recycling	Lack of suitable recycling technologies and/or recycling infrastructure	Lack of price incentives for recycling
✓ Tantalum	+	+	+	-
Indium	+	-	+ -	-
Ruthenium	+	-	+ -	-
Palladium	+ -	-	+ -	-
Platinum	+ -	-	+ -	-
✓ Rare earths	+ -	+	+	-
Gallium	+	-	+ -	-
Tellurium	+	-	-	-
Germanium	+	-	+ -	-
Cobalt	+ -	-	-	-
✓ Lithium	+ -	+	+	+

Figure 7.6: Prioritization step 4: Focused distinction of recycling restrictions

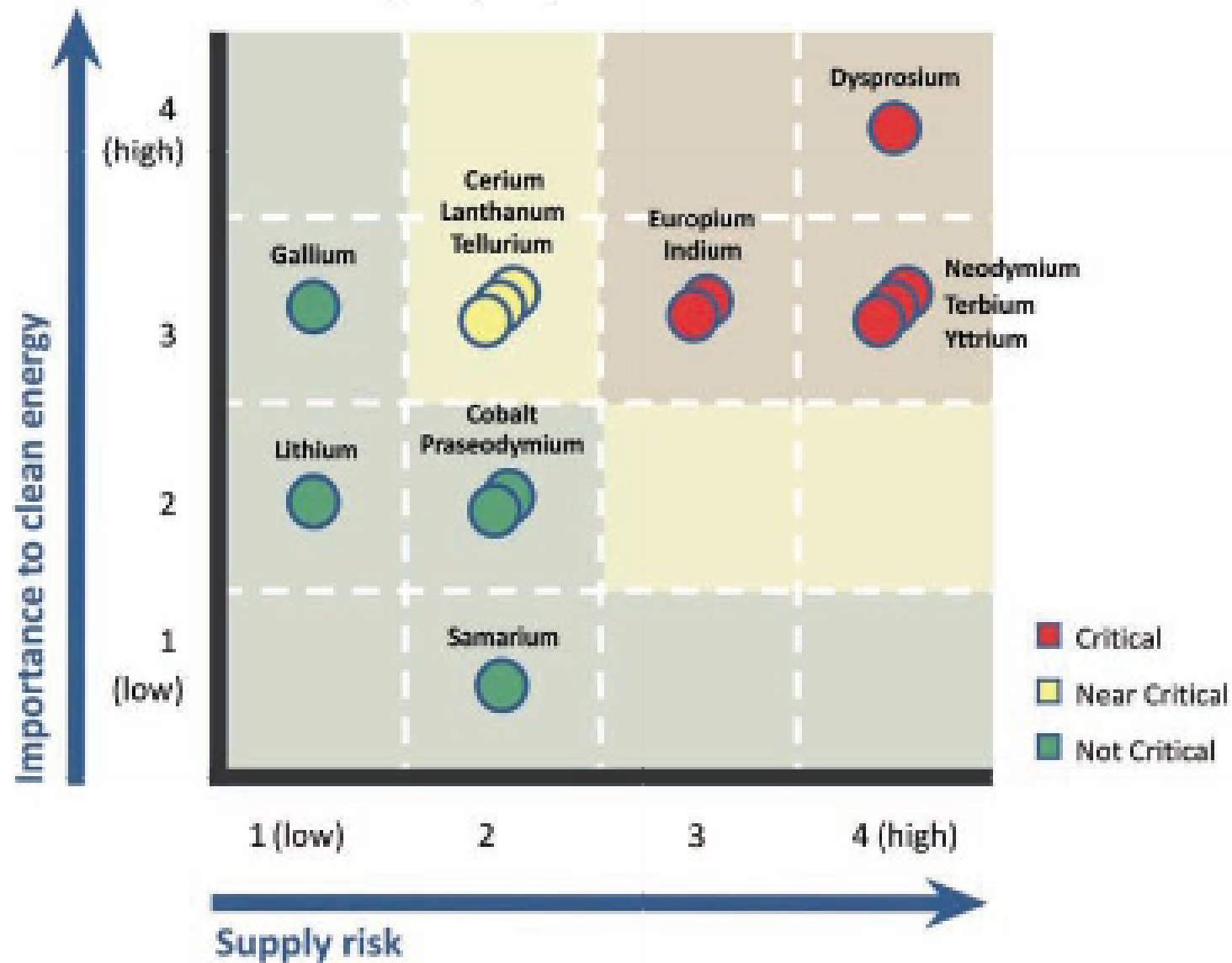
timeline	Metal
<b>short-term (within next 5 years)</b> + rapid demand growth + serious supply risks + moderate recycling restrictions	Tellurium Indium Gallium
<b>mid-term (till 2020)</b> + rapid demand growth and + serious recycling restrictions  <u>or:</u> + moderate supply risks + moderate recycling restrictions	Rare earths Lithium Tantalum  Palladium Platinum Ruthenium
<b>long-term (till 2050)</b> + moderate demand growth + moderate supply risks + moderate recycling restrictions	Germanium Cobalt

Figure 7.8: Summarized prioritization regarding timeline of this study

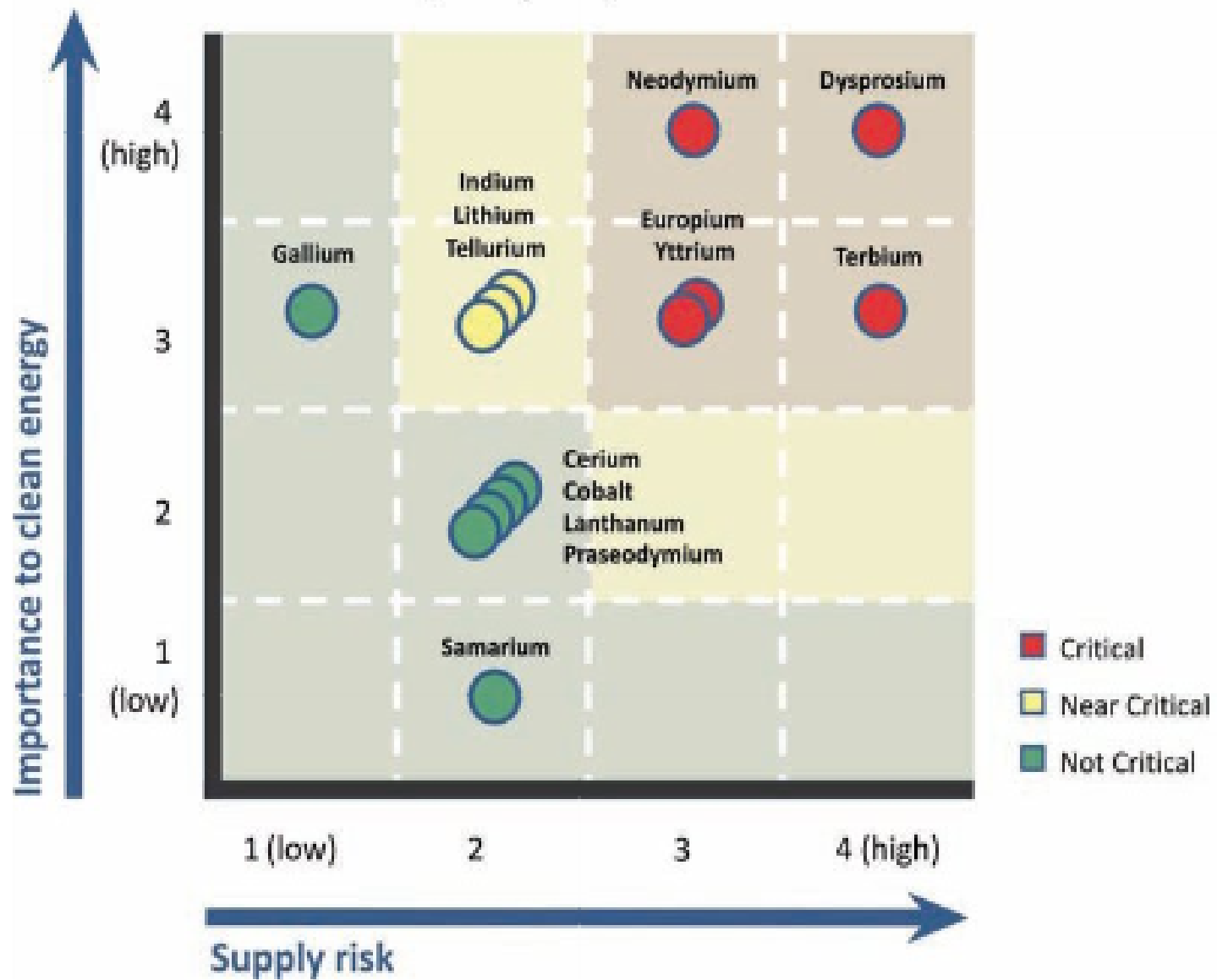
## Energy critical elements as defined by DOE (2011)



### Short Term (0–5 years)



### Medium Term (5–15 years)





## **Critical and strategic minerals will change with time.**

- 1) What is the global demand likely to be?
- 2) What is the US supply likely to be?
- 3) What are the limits and obstacles to US production?
  - a) land access
  - b) permitting speed
- 4) What needs to be done to convert marginal resources into reserves?
- 6) What are the new types of deposits and ore-forming systems of the future?

# **Some of the challenges in producing these technologies**

- How much of these minerals do we need?
- Are there enough materials in the pipeline to meet the demand for these technologies and other uses?
- Can any of these be recycled?
- Are there substitutions that can be used?
- Are these minerals environmental friendly—what are the reclamation challenges?
  - REE and Be are nearly always associated with U and Th and the wastes from mining REE and Be will have to accommodate radioactivity and radon

# Bottlenecks

- Risk and timing of investment
  - Unpredictable
  - Rapid change in demand
  - Engineering/design/production of these products is faster than the exploration/mining/processing
- Extraction
  - Supplies
  - Economically feasible in a timely manner
- Refining
  - Technically feasible
  - Economical

# FUTURE GEOLOGICAL RESEARCH

- Need for understanding the mineralogy and distribution of these minerals in known ore deposits
  - Geologic mapping (lithology, structure, alteration)
  - Geologic deposit models
  - Mineralogy/chemistry
- Are there additional geologic sources for some of these minerals?
- What are the potential environmental consequences of mining these minerals and how do we mitigate them?

# ASSIGNMENT

- **NEXT WEEK (Feb 12) IS Be, REE AFTER THAT (Feb 19, March 5, 12)**
- Barton and Young, S., 2002, Non-pegmatitic deposits of beryllium: mineralogy, geology, phase equilibria and origin: Reviews in Mineralogy and Geochemistry, v. 50, p. 591-691.  
[http://www.geo.arizona.edu/~mdbarton/MDB\\_papers\\_pdf/Barton02\\_BeRiMG050.pdf](http://www.geo.arizona.edu/~mdbarton/MDB_papers_pdf/Barton02_BeRiMG050.pdf)
- McLemore, 2010, NMBGMR OF 533,  
[http://geoinfo.nmt.edu/publications/openfile/downloads/OFR500-599/526-550/533/ofr\\_533.pdf](http://geoinfo.nmt.edu/publications/openfile/downloads/OFR500-599/526-550/533/ofr_533.pdf)